

REINHOLD ENVIRONMENTAL Ltd.



**2018 NO<sub>x</sub>-Combustion Round Table  
& Expo Presentation**

February 19-20, 2018, in St. Louis, MO / Hosted by Dynegy

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# Mercury Mitigation – Update on Compliance Strategies and Technologies

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# Training Road Map

- Coal basics
- Mercury from Coal to Stack
- Overview Mercury Control Strategies
  - Technology basics
  - Review of strategies used for MATs compliance

# Federal Mercury and Air Toxics Standards (MATS)

## ▶ What's regulated?

▶ Mercury (Hg), acid gases (HCl) and particulate matter (PM)

## ▶ Who's affected?

▶ All coal-fired electricity generating units (EGUs) greater than 25 MW electricity

## ▶ When?

▶ Almost all compliance by April 16, 2016



## Update on MATS

- In April 2017, a review by the U.S. Court of Appeals for the D.C. Circuit of a “supplemental finding” associated with the cost benefit analysis of the MATS rule conducted by EPA was stayed at the request of the Trump Administration. The court case continues to be stayed indefinitely, but the MATS Rule is still in force until the issue is resolved. The EPA has not issued any further official proceedings regarding MATS rulemaking

# Primary Factors that Affect Mercury Emissions and Control Strategies

- Coal properties
  - Mercury, halogens, sulfur
- Air Pollution Controls
  - NO<sub>x</sub>, SO<sub>x</sub>, PM
- Other Considerations for Supplemental Controls
  - Economics
  - Balance-of-plant impacts
  - Environmental constraints (e.g. ancillary emissions or effluent discharges)

# What Does It All Mean?

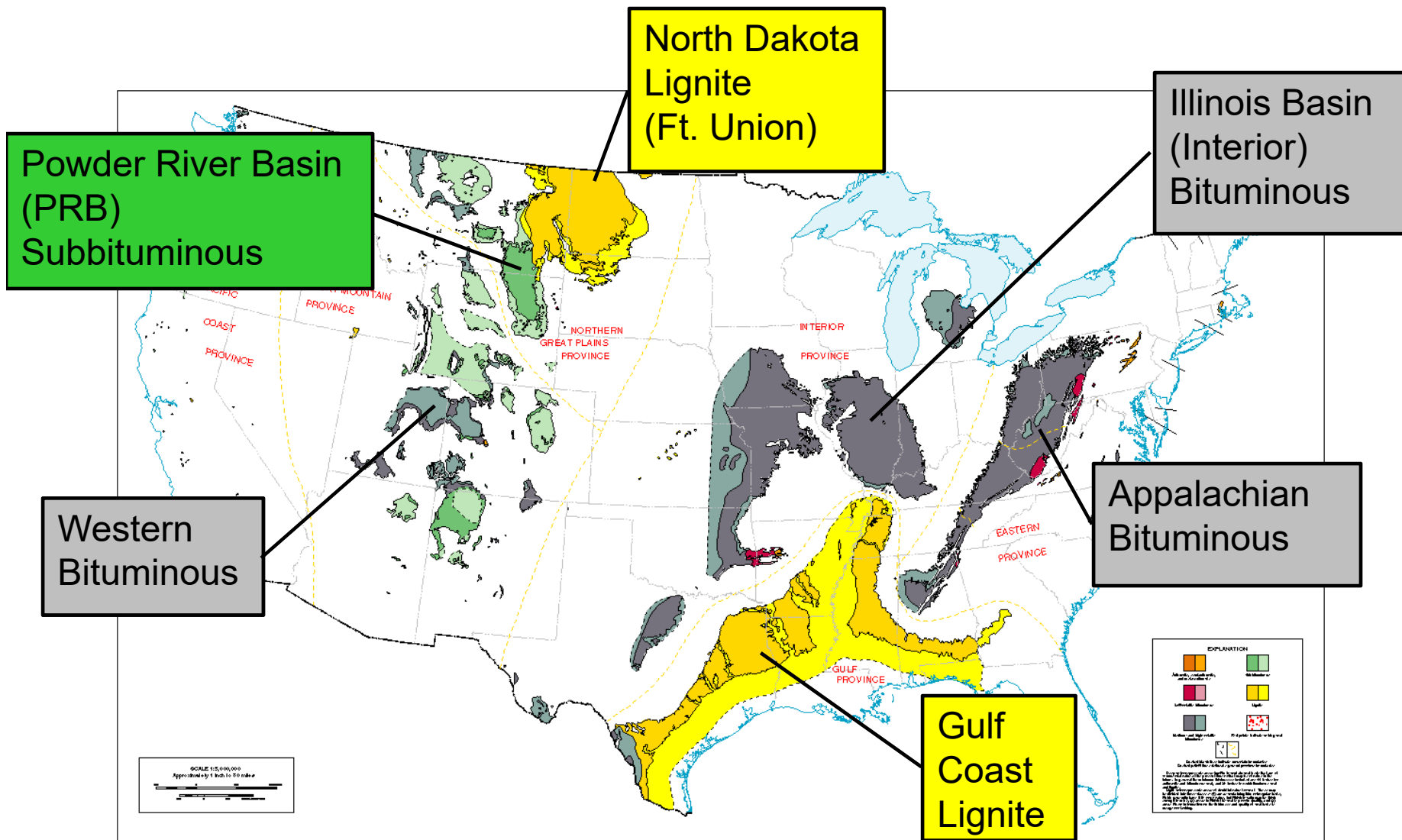
- ▶ Many plants have relatively new processes for controlling one or more of these air emissions:
  - ▶ Hg, HCl, SO<sub>2</sub>, SO<sub>3</sub>
- ▶ Plus all the previous air pollution control!
- ▶ We can learn what combinations of pollution control work by coal type by reviewing what the industry is using to meet compliance



# Coal Properties



# Major Coal Fields in the US



# Typical Coal Properties



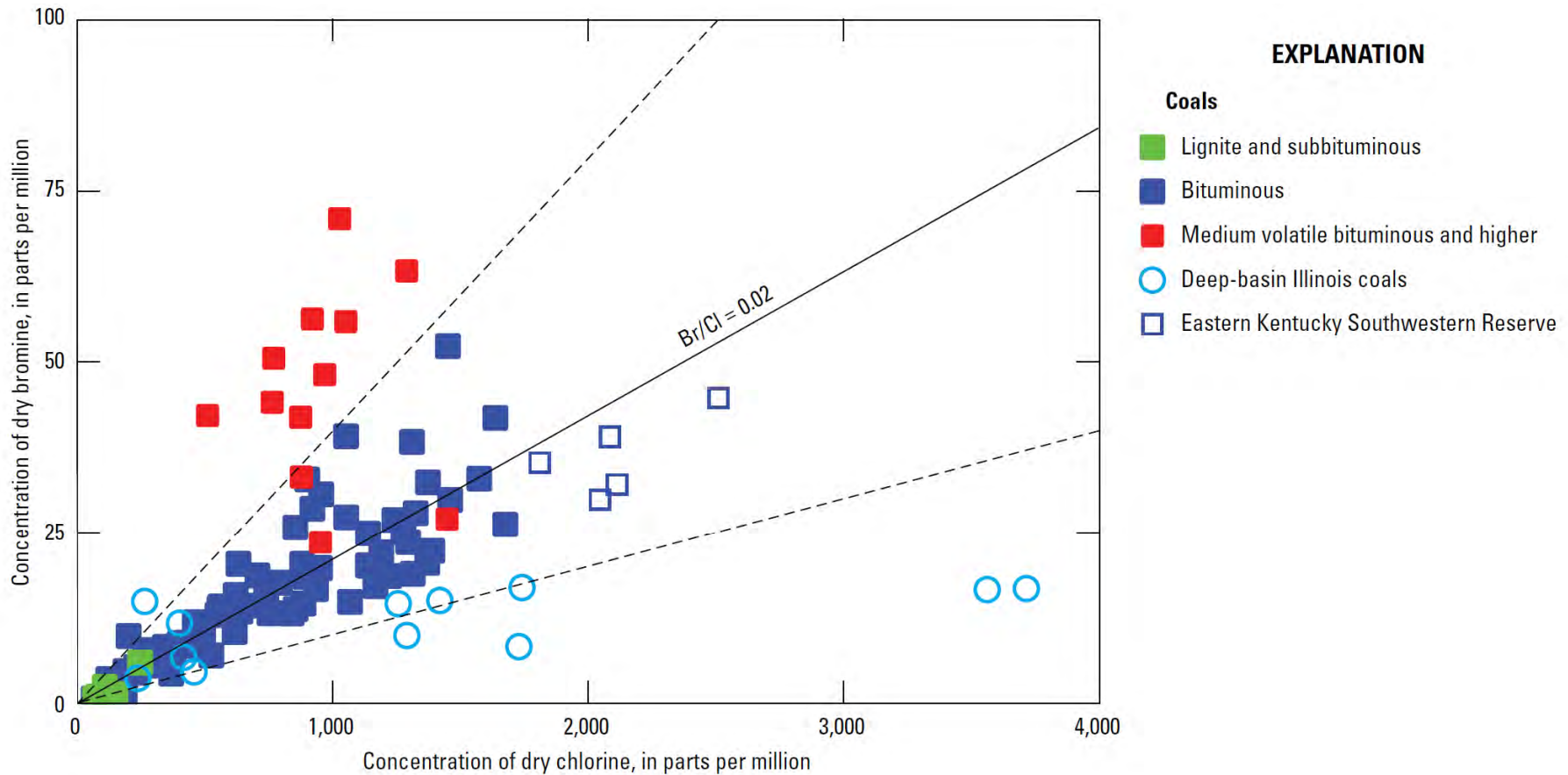
Rank	Lignite	Subbituminous	Bituminous	Anthracite
Age (million yrs)	~60	~100	~300	~350
Heating Value	~ 7,000	~ 10,000	12,000-15,000	~ 15,000
% Carbon	65-72	72-76	76-90	90-95
% Moisture	30-70	10-30	5-10	~5
%Sulfur*	0.4-1	<2	1-4	0.6-0.8
Chlorine* (ppm)	50-150	<100	100-2000+	
Mercury* (lb/TBtu)	Gulf: 20-40 Ft. Union: 10-30	PRB: 5-15	NAPP: 15-20 CAPP: 5-15 East Int: 5-10	PA: 5-10

\* Key contributors to Hg emissions

Mercury from USGS: <https://pubs.usgs.gov/fs/fs095-01/fs095-01.html>

Chlorine from USGS:

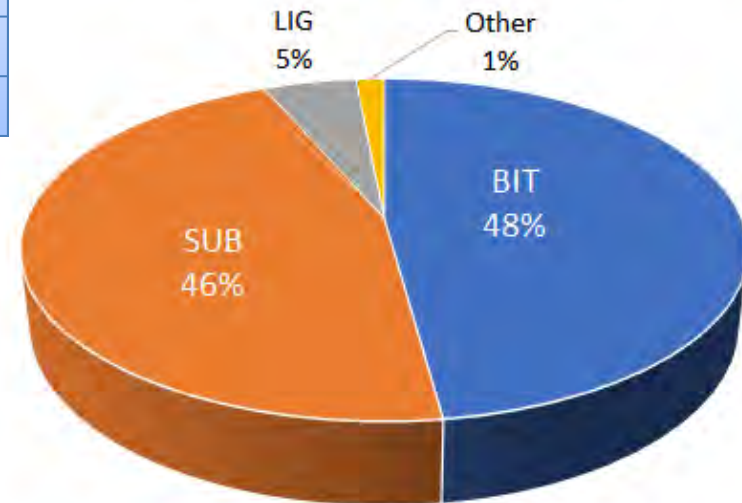
# Coal Halogen: Chlorine and Bromine



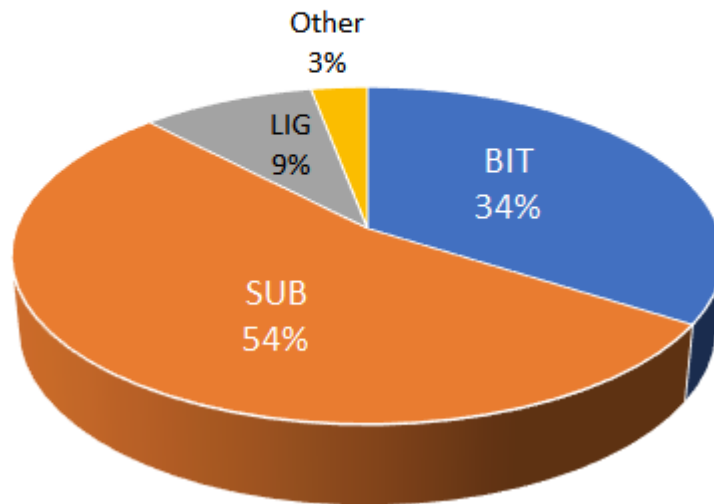
# Primary Coal-Type Fired in 2017

Coal	MW	Coal (MMtpy)	Avg CF (%)	EGUs
BIT	125,481	226.5	43%	294
SUB	119,888	358.5	55%	294
LIG	12,928	61.3	69%	29
Other	3,944	19.9	71%	26

**Installed Capacity (of 262 GW)**



**Coal Fired (Dec 16-Nov 17)  
Percent of 666.2 MMtons**

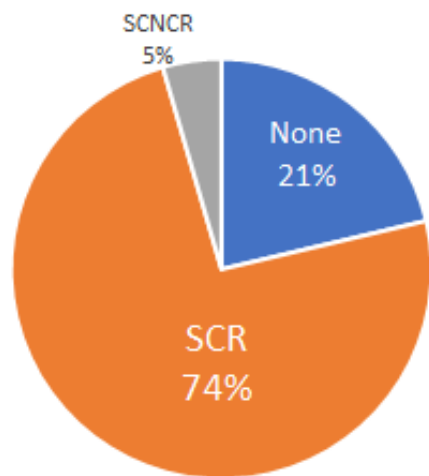
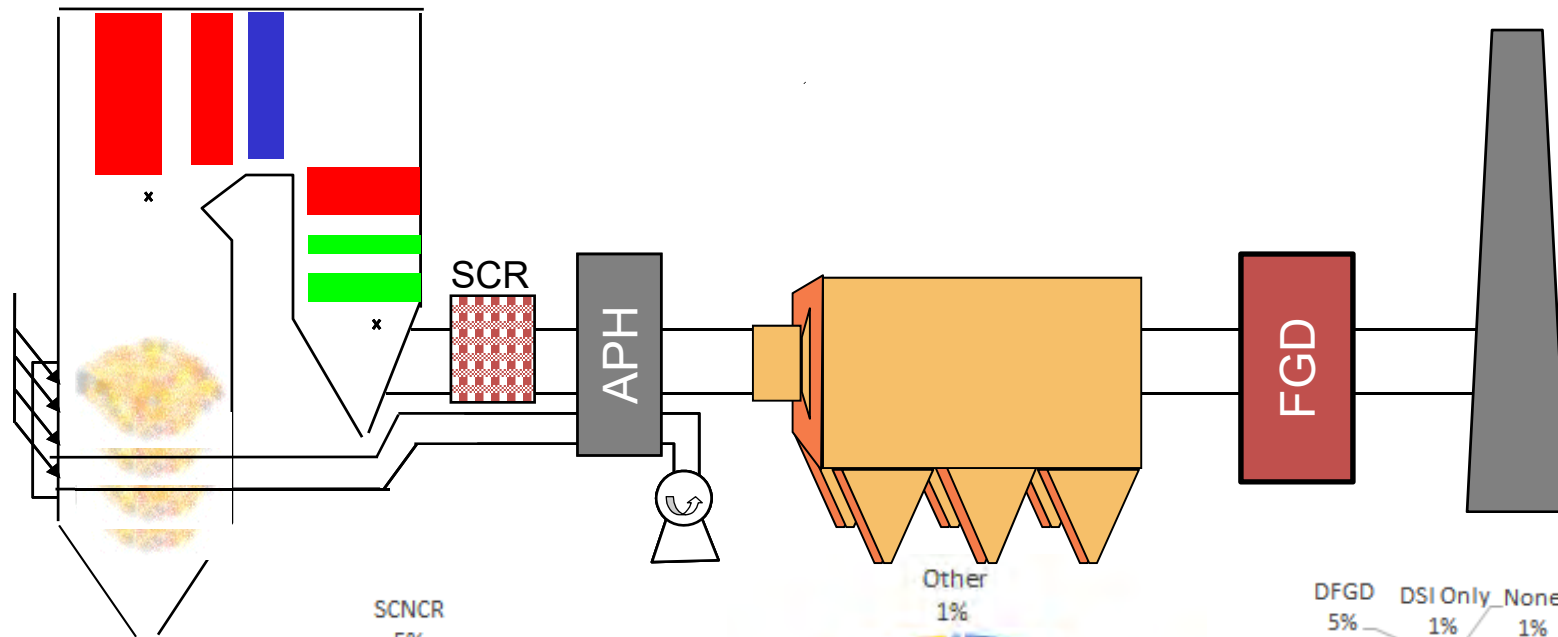


- The fraction of installed capacity firing primarily bituminous and subbituminous is similar
- 54% of coal fired for generation is subbituminous
  - Higher capacity factor (CF) for subbituminous-fired units

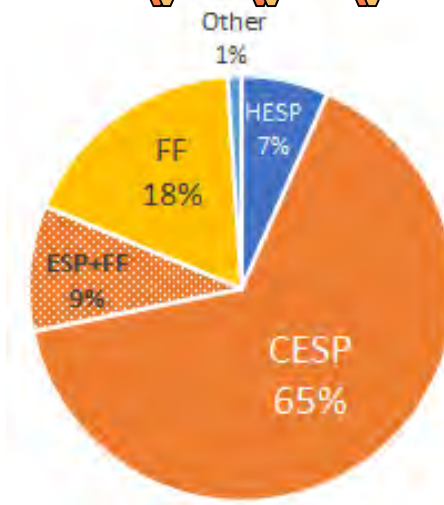
# Air Pollution Controls



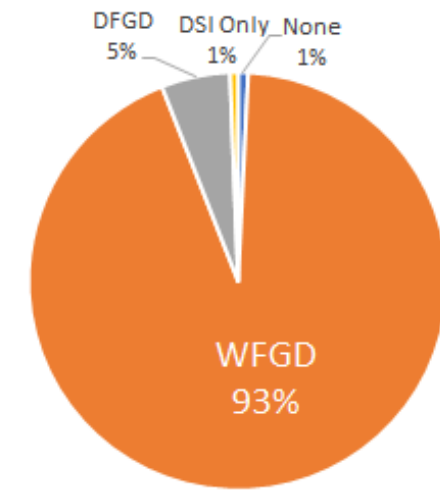
# APC Configurations: Bituminous Plants



NO<sub>x</sub>

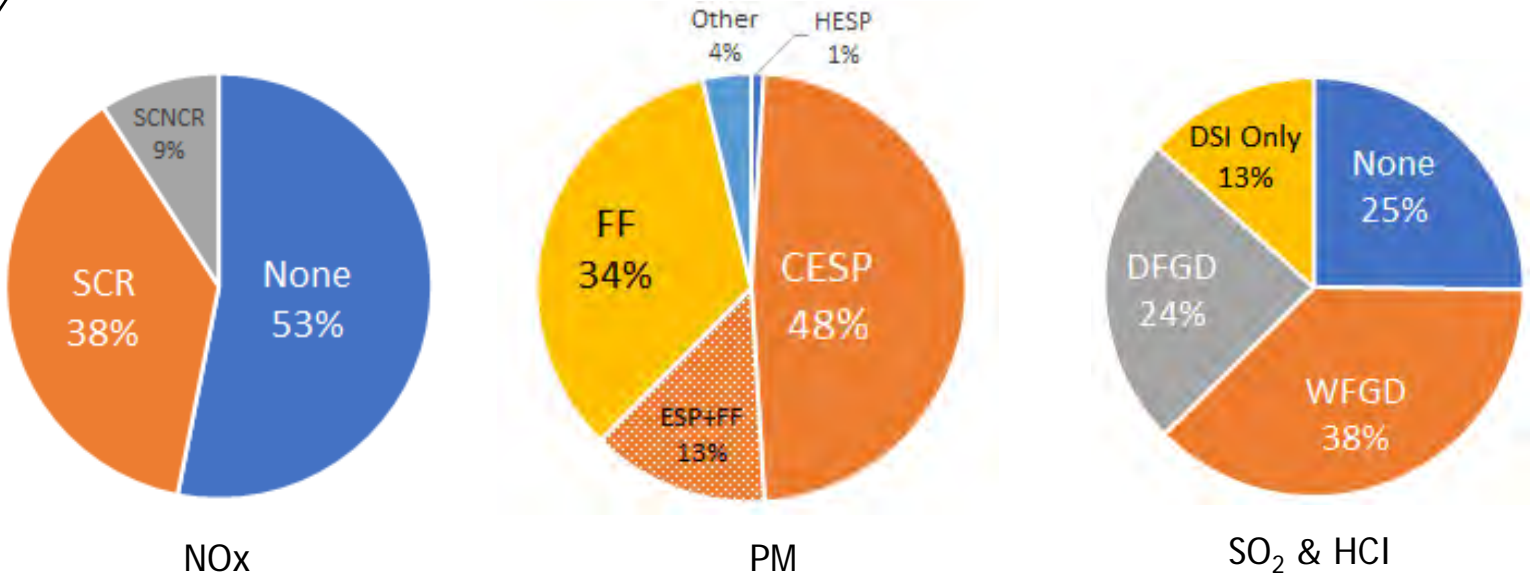
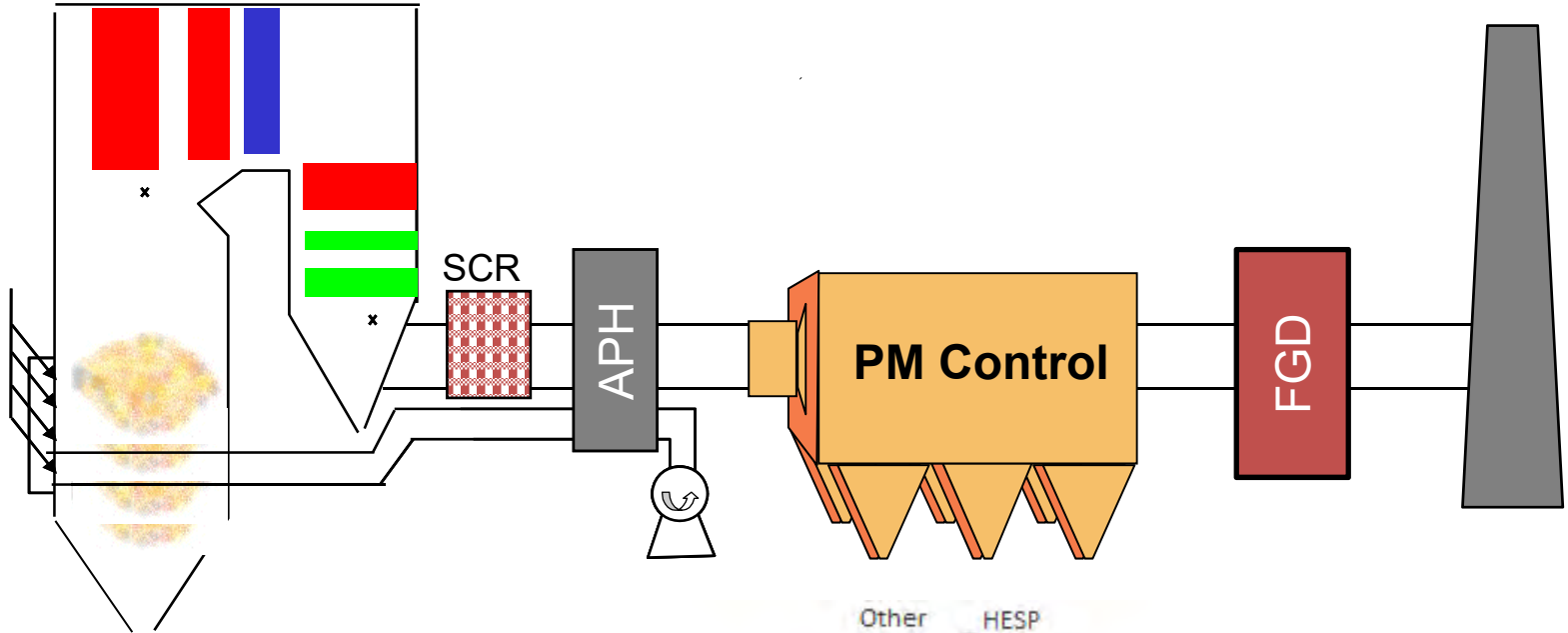


PM



SO<sub>2</sub>

# APC Configurations: Subbituminous Plants



# Mercury in Coal Plants

- Fundamental Chemistry
- Control Strategies

# Mercury: What Do We Mean by Speciation?

## ➤ Gas-phase mercury:

- Elemental:  $\text{Hg}^0$
- Oxidized:  $\text{Hg}^{+2}$  ( $\text{HgCl}_2$ ,  $\text{HgBr}_2$  other species?)

Oxidized mercury is **VERY** soluble in water (120°F)

$\text{HgCl}_2$ : **13.2g/100g water**

$\text{HgBr}_2$ : **1.3g/100g water**

$\text{Hg}^0$ : **< 1 x 10<sup>-5</sup>/100g water**

## ➤ Particulate mercury:

- $\text{Hg}_p$
- Mercury (adsorbed on particles)

# Two Ways to Remove Mercury

## Remove with PM

### Adsorb Hg on particles

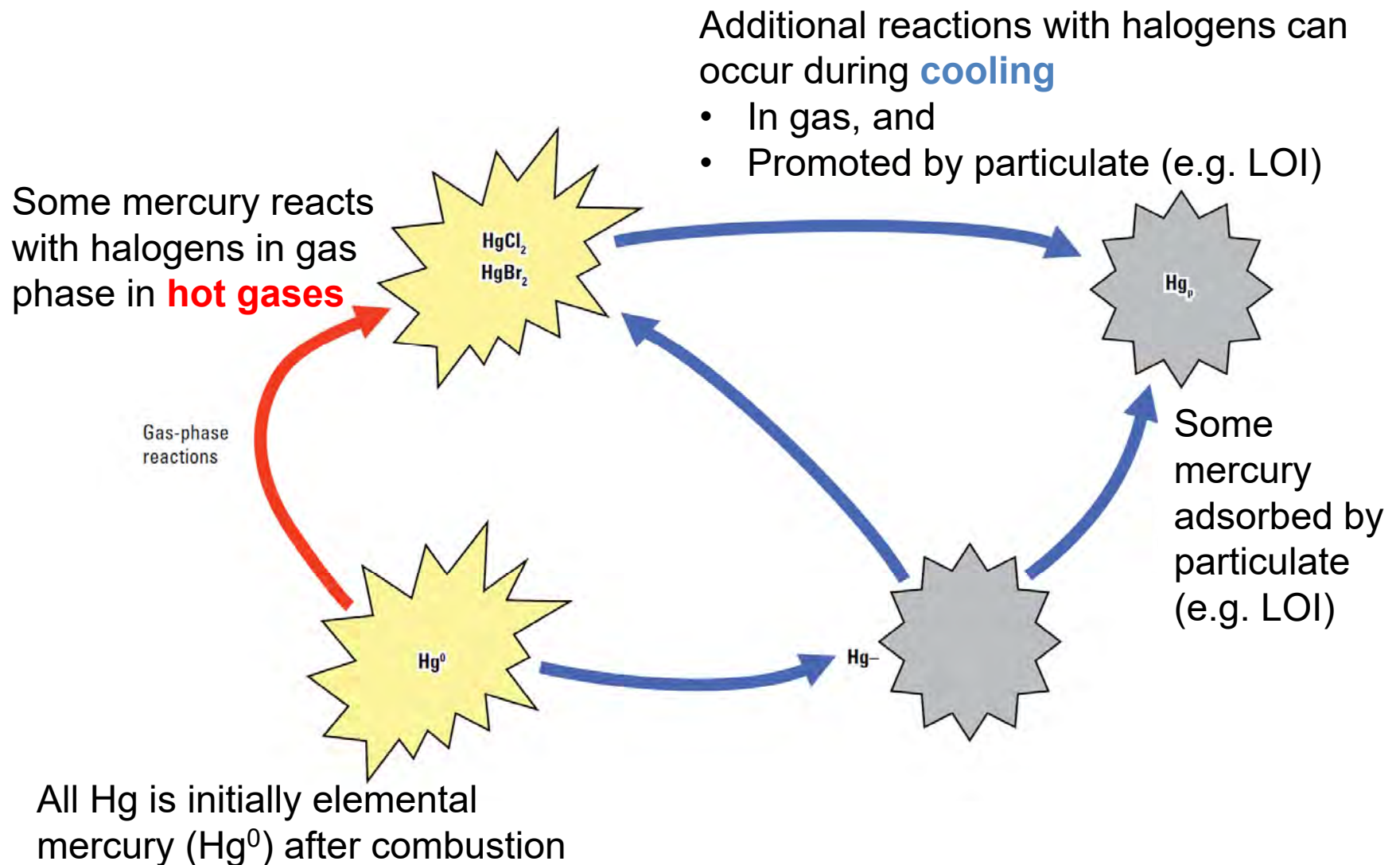
- Unburned carbon in fly ash
- Sorbent injection
- Fixed adsorption structures

## Remove in scrubber

### Absorb Hg (Primarily Hg<sup>2+</sup>)

- Wet flue gas desulfurization (FGD) scrubbers
- Dry FGD scrubbers

# Mercury: Reaction Pathways During Combustion and Cooling



# Mercury Control Strategies



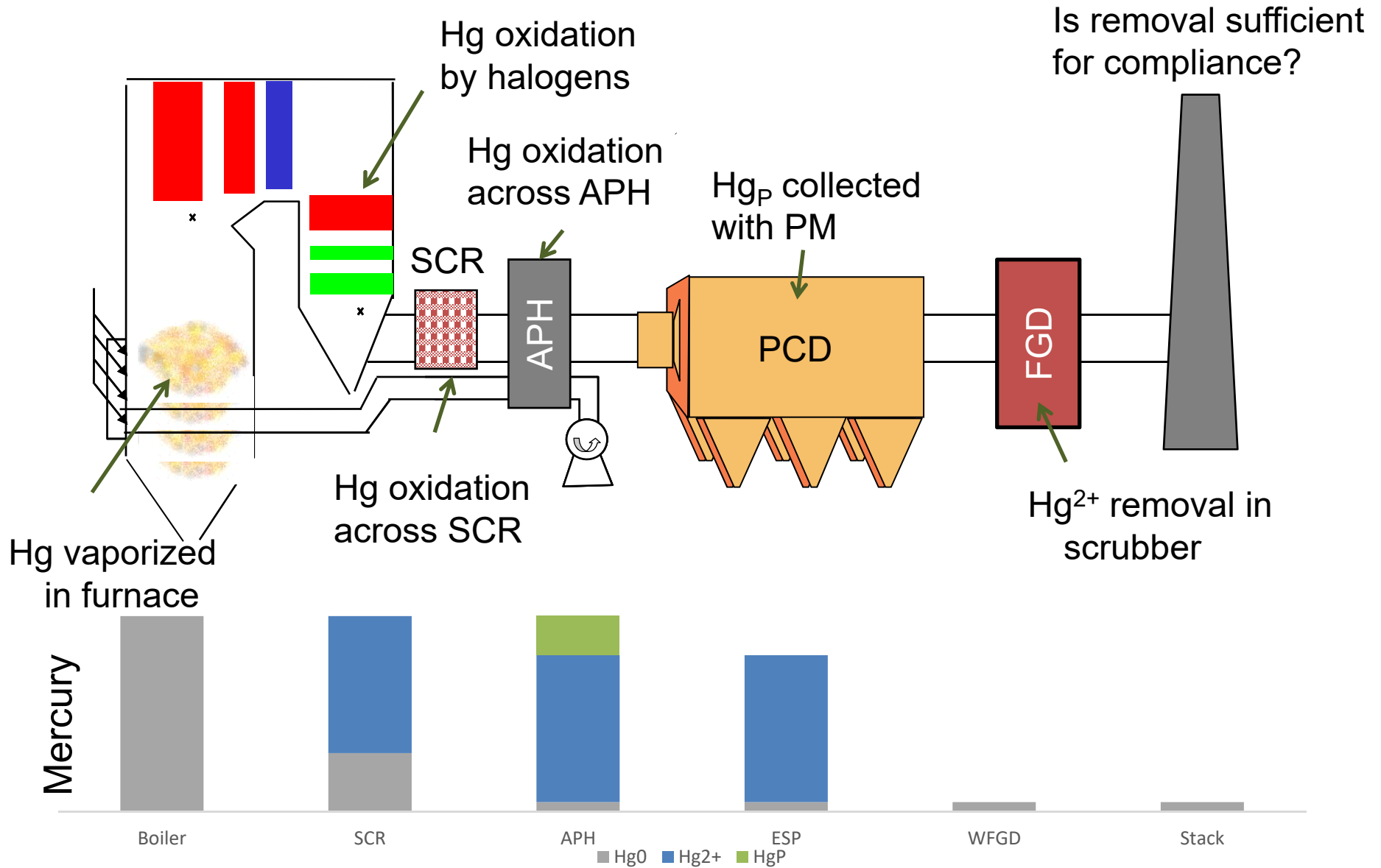
## ➤ Use sorbents

- Powdered Activated Carbon (PAC)
- Brominated Activated Carbon (Br-PAC)
- Halogen or catalyst + PAC
- Use sulfur-tolerant PAC and/or DSI to mitigate  $\text{SO}_3$  as needed
  - Caution: Don't lower  $\text{SO}_3$  too far – it will affect ash resistivity and ESP performance
  - Caution: DSI will increase particulate load to ESP

## ➤ Increase “native” Hg capture

- Enhance Hg removal with unburned carbon through combustion modifications
- Blend fuels to enhance mercury control effectiveness
- Use oxidizing agents or catalysts to increase oxidized Hg

# Mercury Control: "Typical" Bituminous Plant



# Can Plants Rely on Co-Benefits Alone?

## Success Factors

### ➤ Coal and combustion

- Mercury, halogen, sulfur, LOI

### ➤ SCR

- Lower = better: temperature,  $\text{NH}_3$ , age, gas flow rate,  $\text{CO}$ ,  $\text{H}_2\text{O}$ ,  $\text{SO}_2$
- Higher = better: halogen concentration,  $\text{O}_2$
- Other: SCR type, SCR management scheme

### ➤ Particulate Controls

- Hg removed before WFGD: fraction of particulate-phase Hg (LOI, temperature,  $\text{SO}_3$ , ESP SCA, FF cleaning)

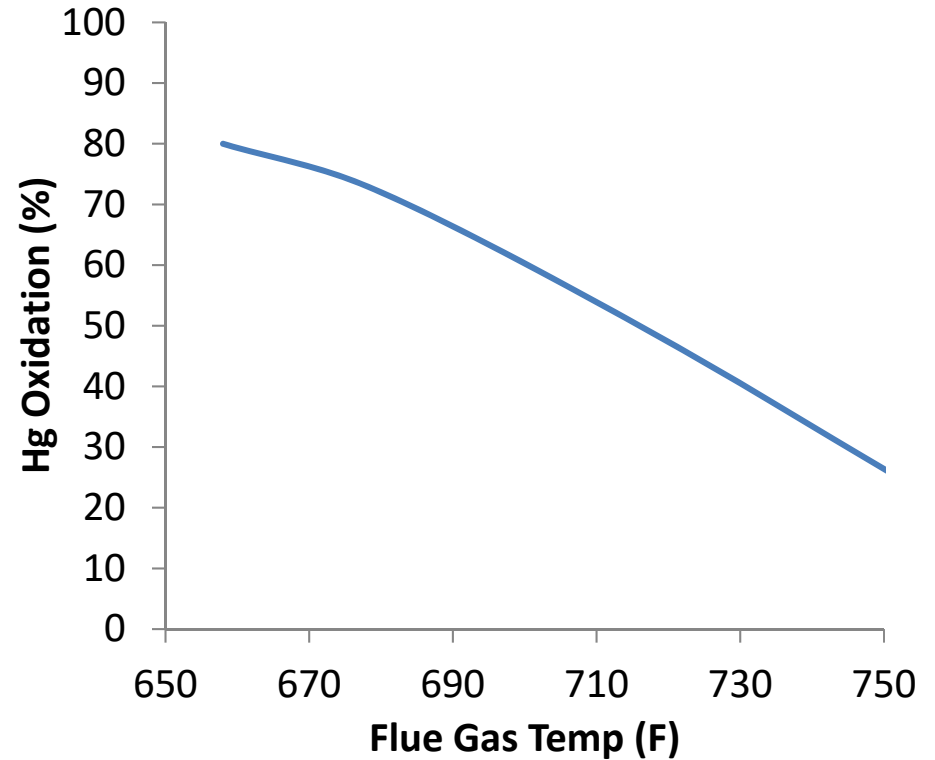
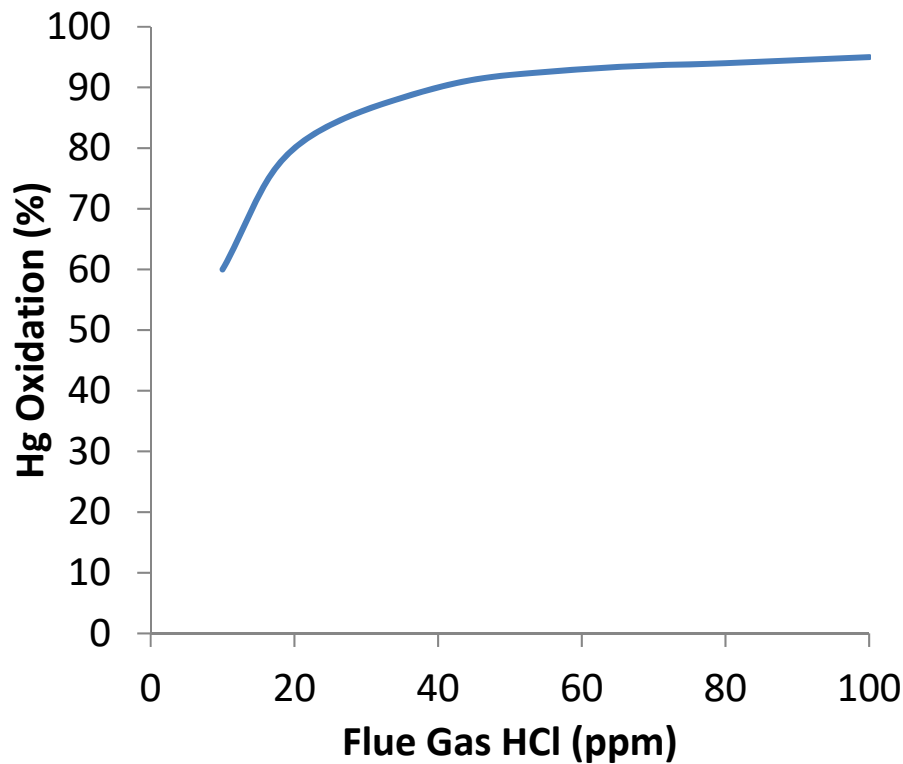
### ➤ Scrubber

- Fraction of oxidized Hg at inlet, ORP, halogens, temperature, pH

#### **YES:**

35% of bituminous units and  
5% of subbituminous units  
rely on co-benefits alone

# Factors Affecting Hg Oxidation Across SCRs



Higher temperature → Lower oxidation

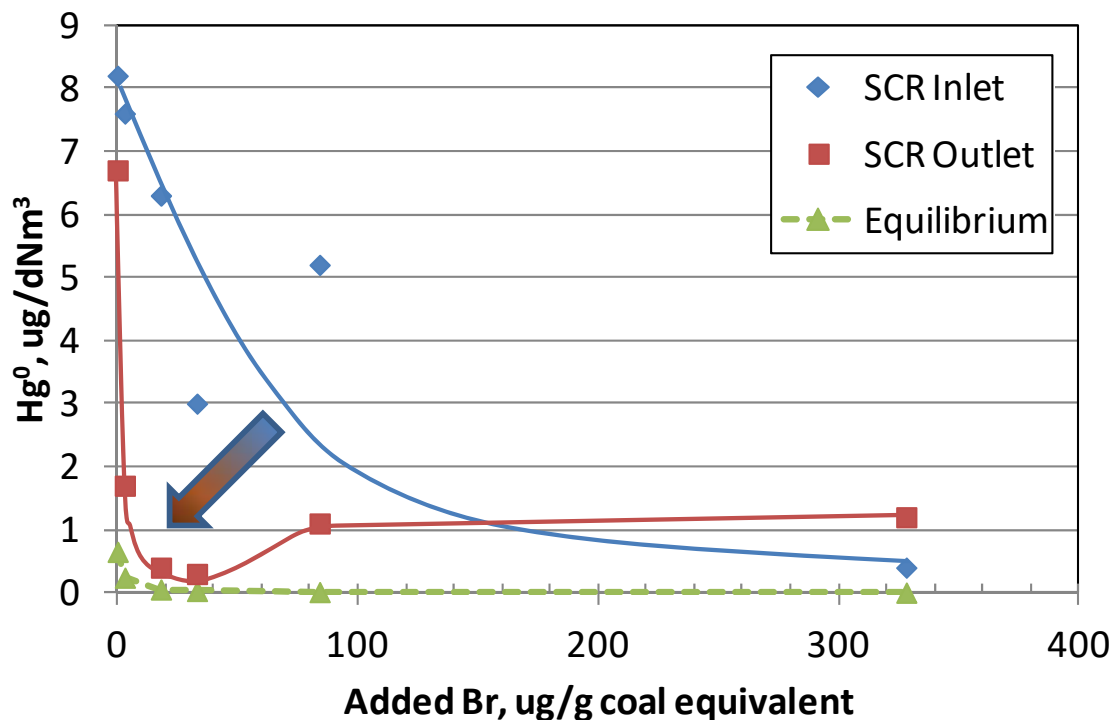
Higher ammonia → Lower oxidation

Review of emissions data from several bituminous plants with SCR+WFGD do not show increased emissions during summer months

Shintaro Honjo, Mitsubishi Heavy Industries America, Mega Symposium 2012

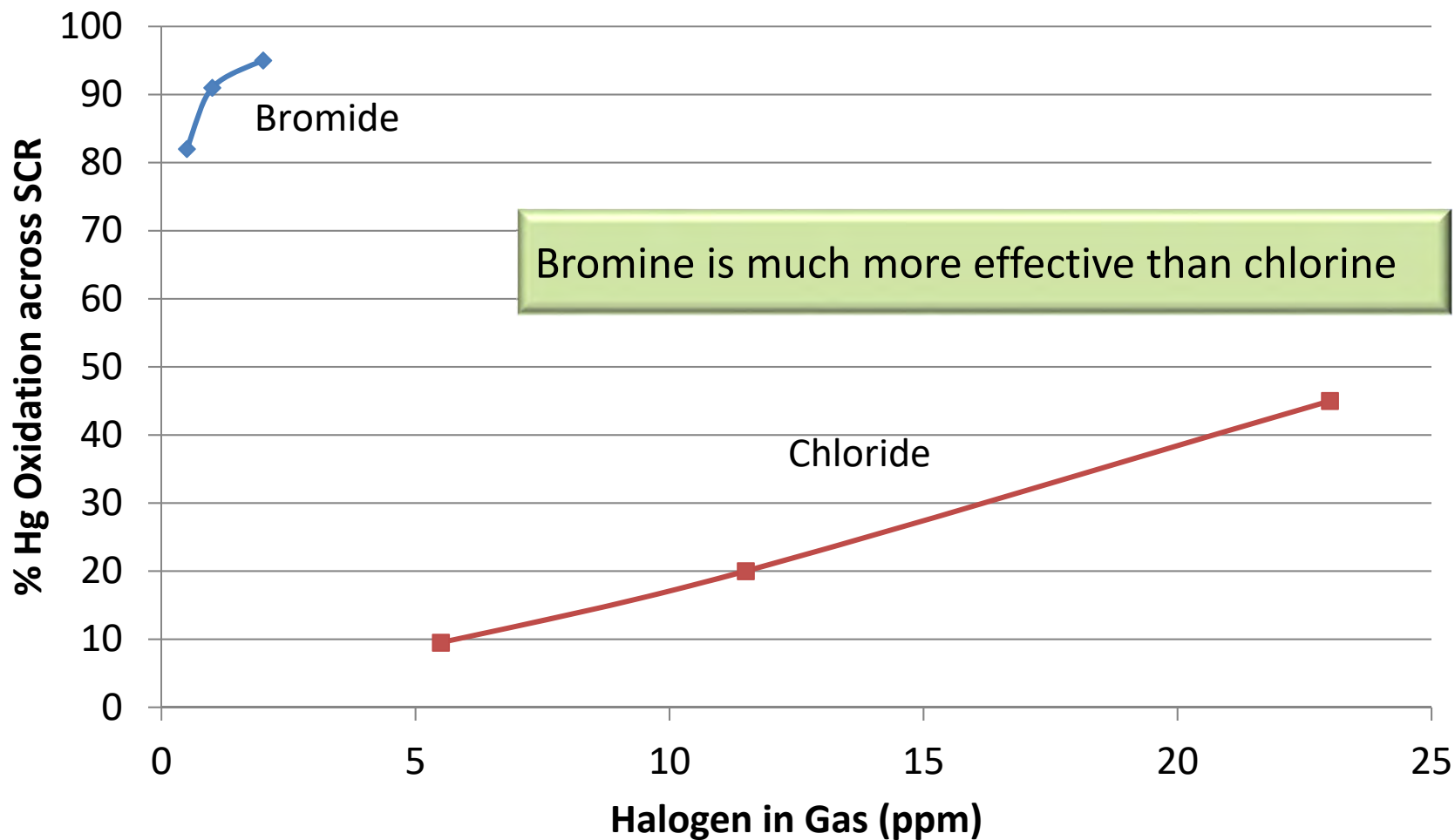
## Effect of Added Halogens on Mercury

- Halogens (Cl, Br, etc.) promote oxidation of Hg in the boiler and SCRs improve this oxidation
- Adding Br to the fuel makes the shift to  $\text{Hg}^{2+}$  take place at higher temperatures (than Cl)



Comparison of Plant Miller data of Berry *et al.* with calculated equilibrium

# Improving Hg Oxidation Across SCR with Halogens

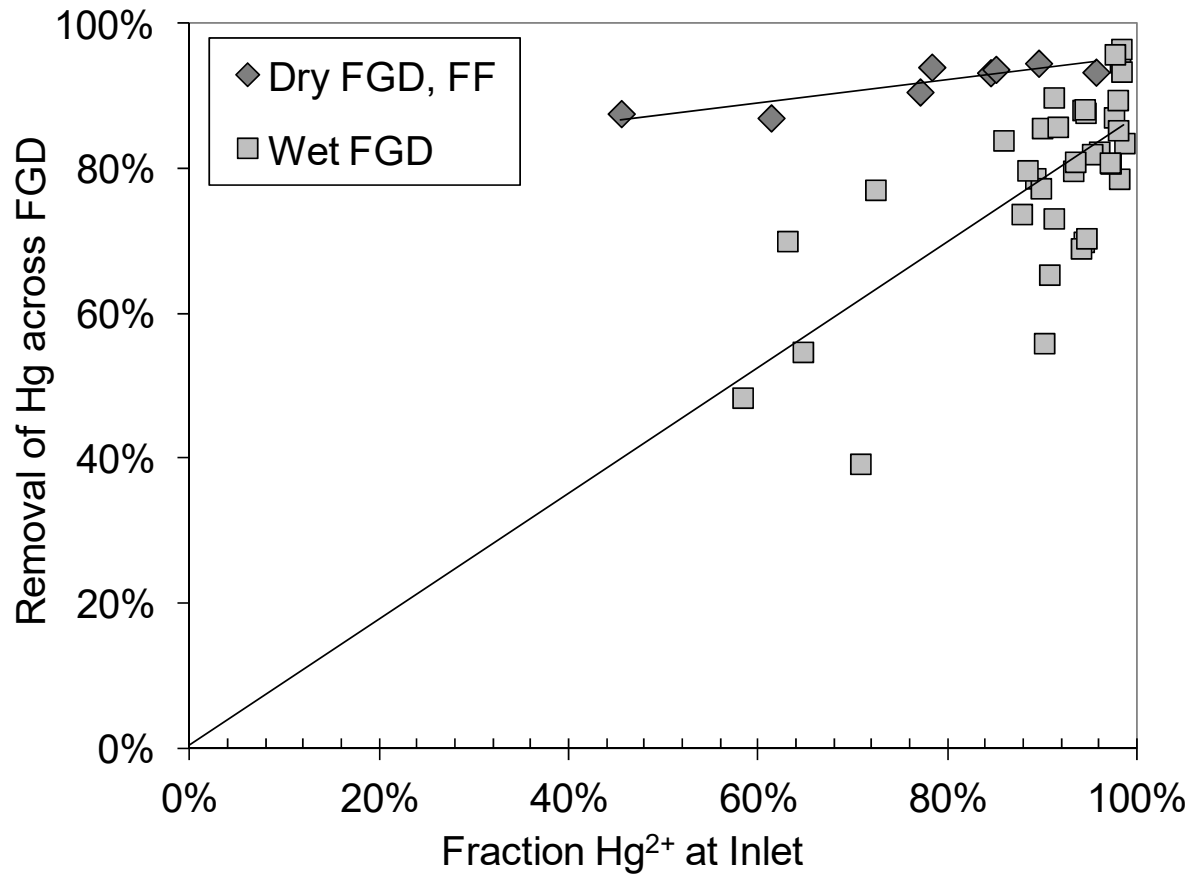


Adapted from Cormetech,  
2015 Reinhold NOx conference



# Benefits of Oxidized Mercury

Many scrubbed plants effectively remove oxidized Hg ( $\text{Hg}^{2+}$ )



## Removal of Hg in Wet FGDs

- Maximize gaseous oxidized Hg at scrubber inlet
- Stabilize Hg<sup>2+</sup> in the liquid
  - Control redox potential (e.g., Mitsubishi Heavy Industries has a patent covering ORP control to optimize net mercury capture)
  - Halogens in the scrubbing solution can complex with Hg<sup>2+</sup> and reduce Hg re-emission (sometimes)
- Increase amount of Hg removed in solid phase
  - Amount of suspended solids in the absorber slurry (impacts fines concentrations and surface area available for mercury adsorption)
  - Iron in fine particles (fines) in the scrubber (from limestone and/or fly ash) that react with Hg
  - Use an additive to the scrubbing solution to tie up Hg or precipitate to solid phase

# Scrubber Mercury Re-emission Additives

- Chemical additives that have successfully improved wet FGD mercury reduction through control of mercury re-emission and can be grouped into five categories:
  - Inorganic sulfides
  - Organic sulfides
  - Organic compounds containing nitrogen and sulfur
  - Organic compounds containing oxygen and sulfur
  - Low molecular weight sulfur-containing polymers
- Some plants add activated carbon directly into the scrubber liquor to remove mercury

# Balancing Gypsum Quality and Mercury Re-Emissions

## Optimum for Gypsum

- High oxidation air rates  
    ↓ sulfite
- Low pH
- High blowdown to manage halogen levels

## Optimum for Hg Control

- Reduce oxidation air
  - $< \sim 300 \text{ mV}^{1*}$
- Increased sulfite
- Increase pH
- High halogen

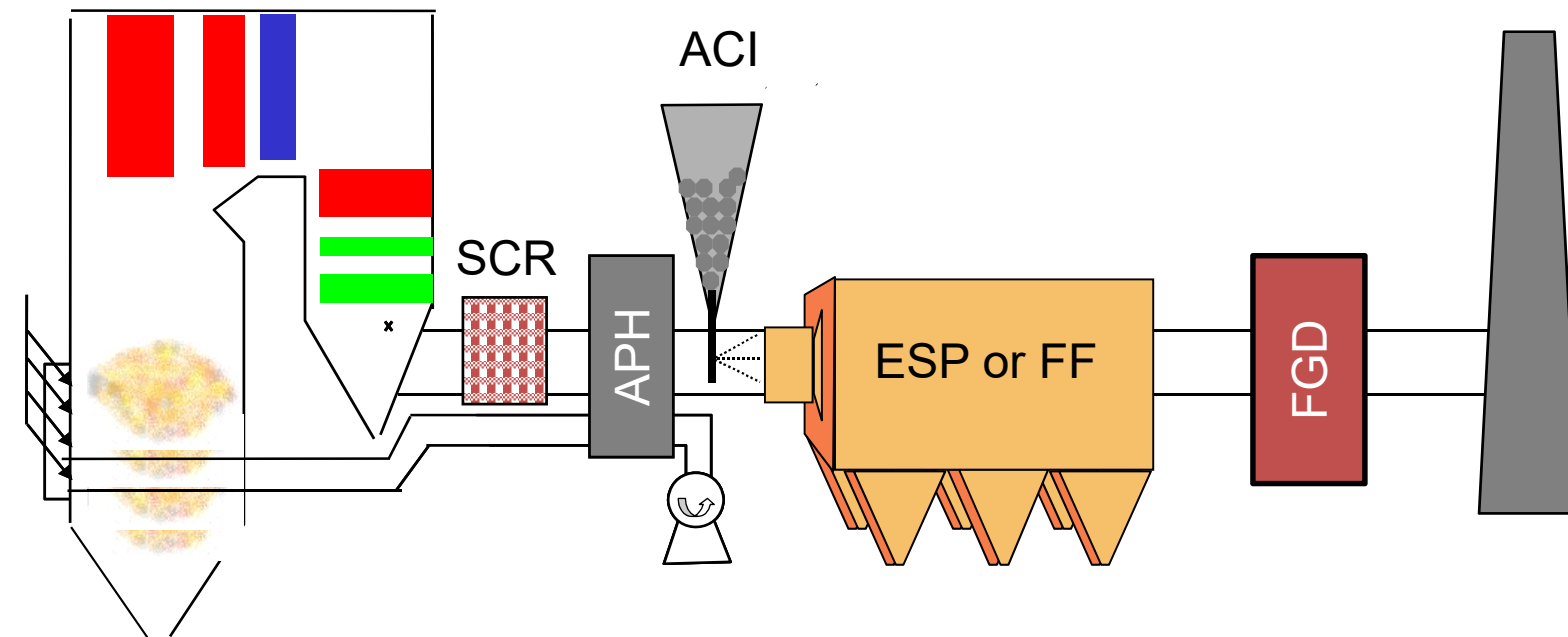


% Dissolved mercury, and risk of re-emission increases at ORP  $> 300 \text{ mv}$

Mandi Richardson, AECOM

# Supplemental Control: Activated Carbon

## Challenges for Hg Removal

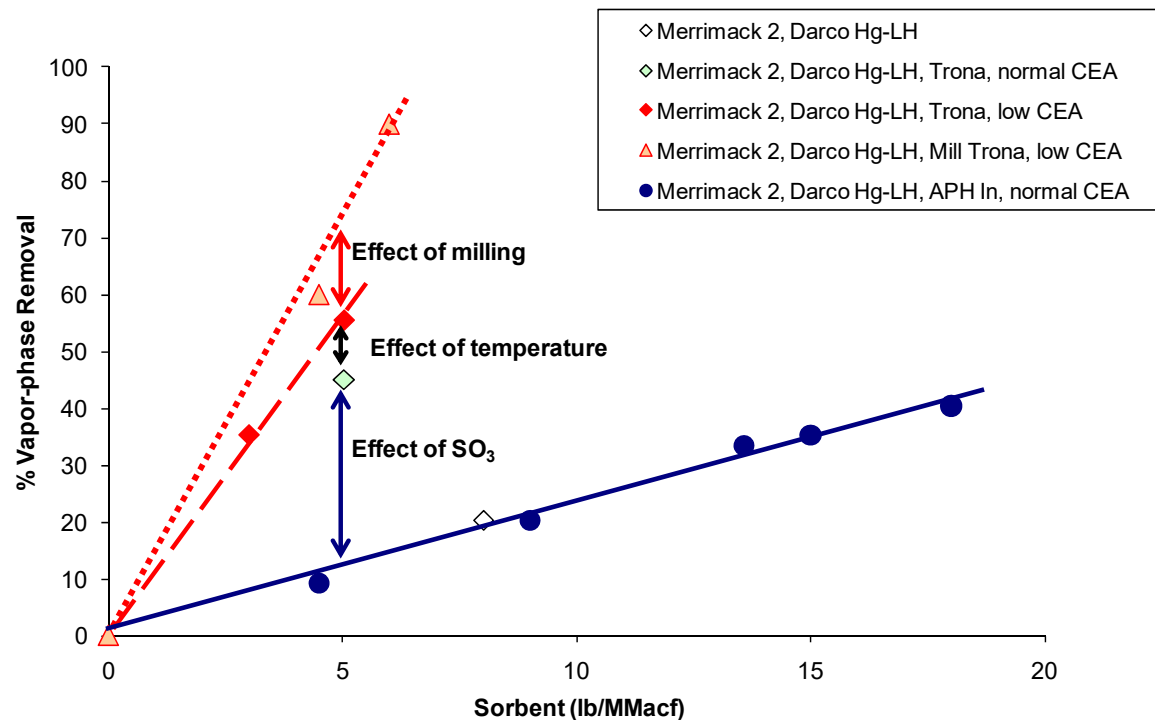


Temperature  
 $SO_3$   
Halogens  
Oxidized Hg

*$SO_3$  and temperature have compounding negative impact on performance of activated carbon*

# Activated Carbon Injection (ACI) with Bituminous Coal-Fired Plants

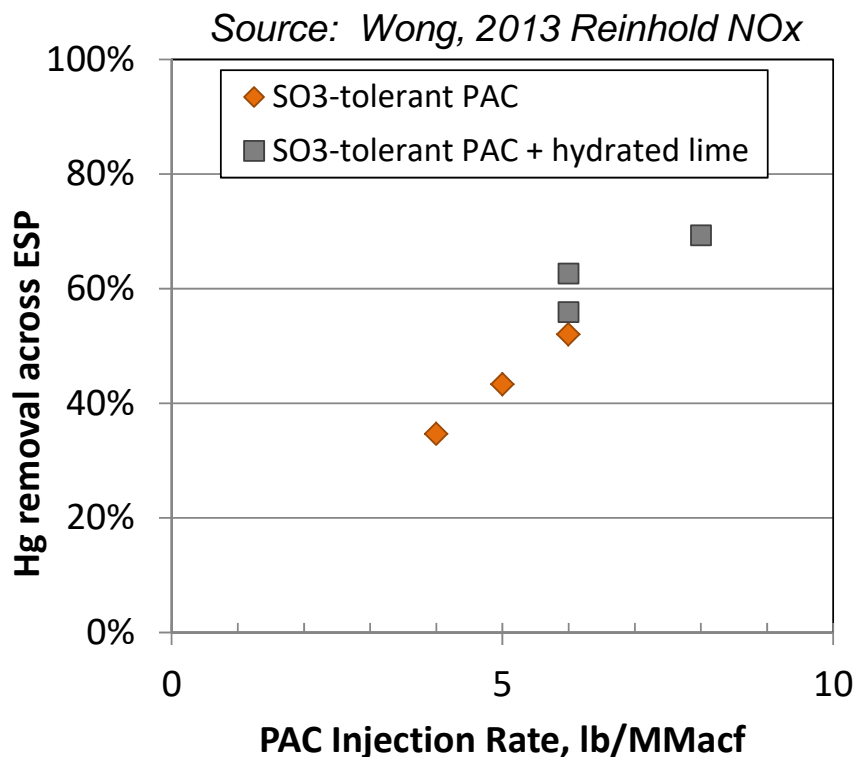
Plants with ESP and high SO<sub>3</sub> (i.e., high-sulfur coals or low-sulfur coals with an SCR/FGD) struggle to achieve high levels of removal



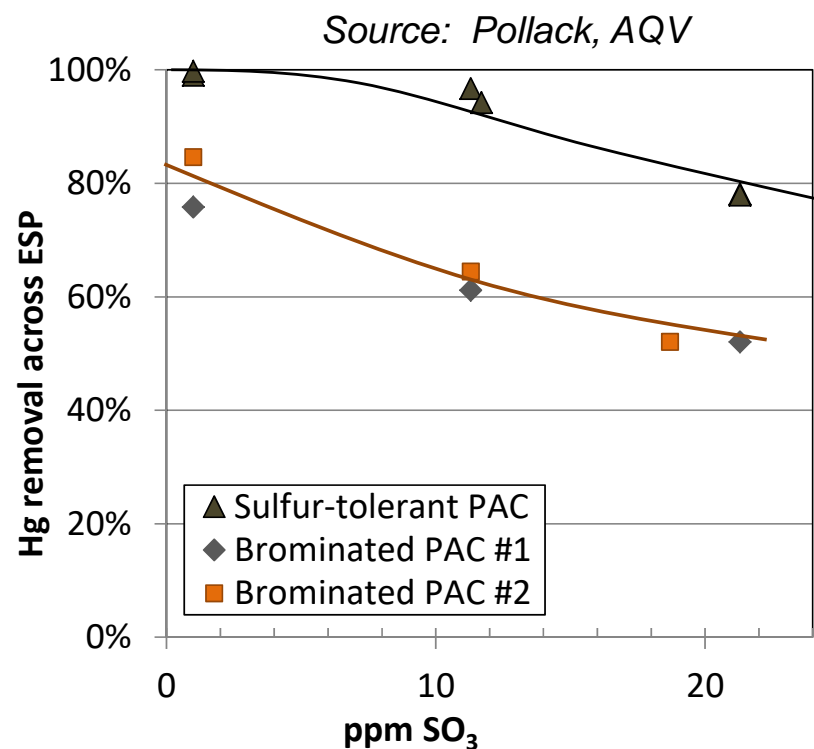
- High temperatures and high SO<sub>3</sub> concentrations in the flue gas downstream of the APH strongly affect sorbent performance
- Other factors, possibly important, including coal chlorine content and sorbent mixing in the flue gas duct

# Sulfur-Tolerant Activated Carbon

Some plants use specialty carbon to achieve mercury control



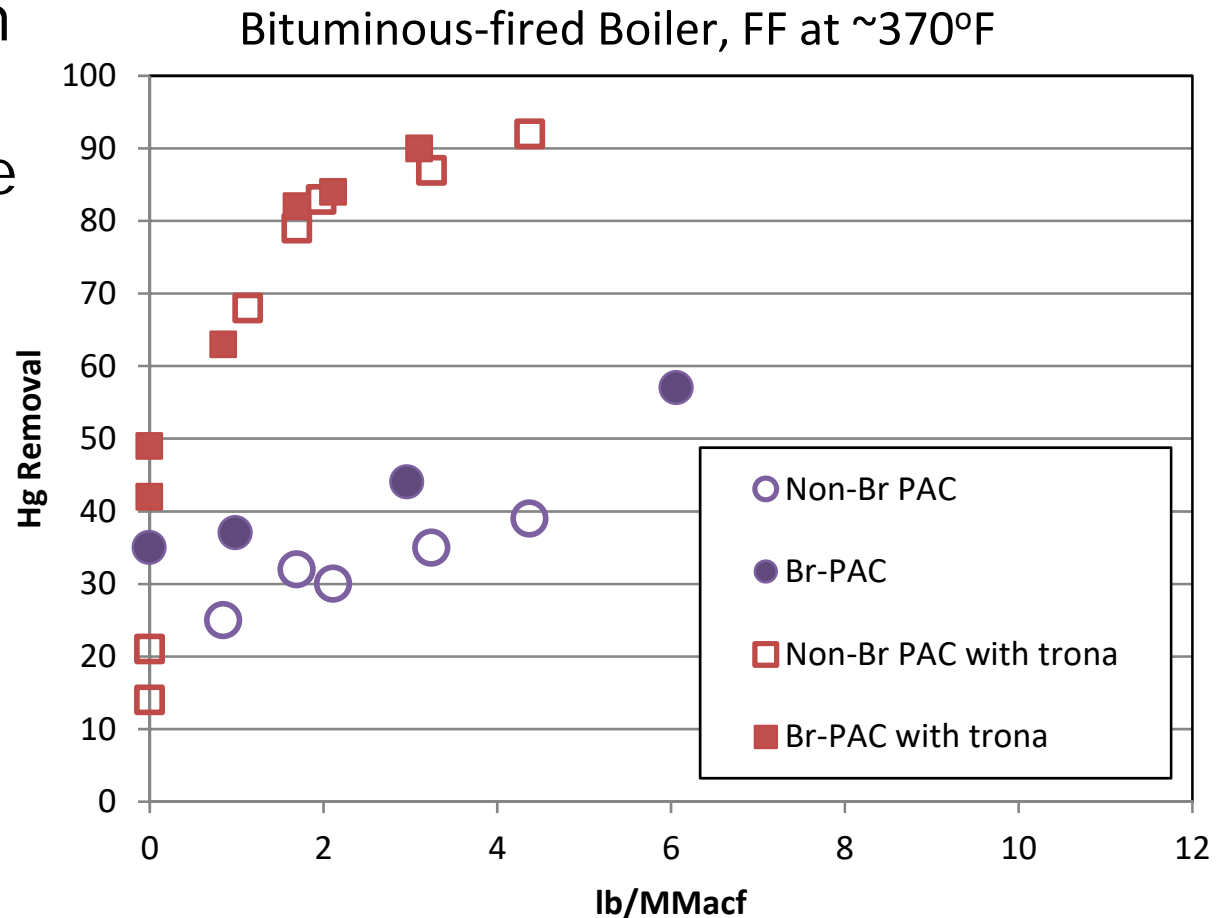
High S coal, SCR-ESP-FGD,  
> 15 ppmv SO<sub>3</sub>



MRC Results: 10 lb/MMacf, injection  
upstream of APH; APH outlet: 300 F

# Variation: Trim with ACI, Use Alkaline Sorbents (DSI) to Lower SO<sub>3</sub>

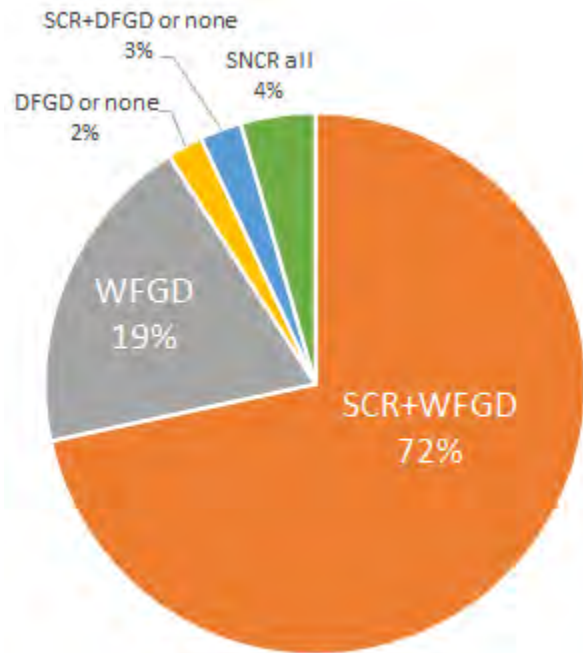
- Sodium or calcium DSI sorbents can be used to remove SO<sub>3</sub> and increase effectiveness of PAC
- Example:  
Bituminous-fired boiler with FF, ~20 ppm SO<sub>3</sub> uncontrolled



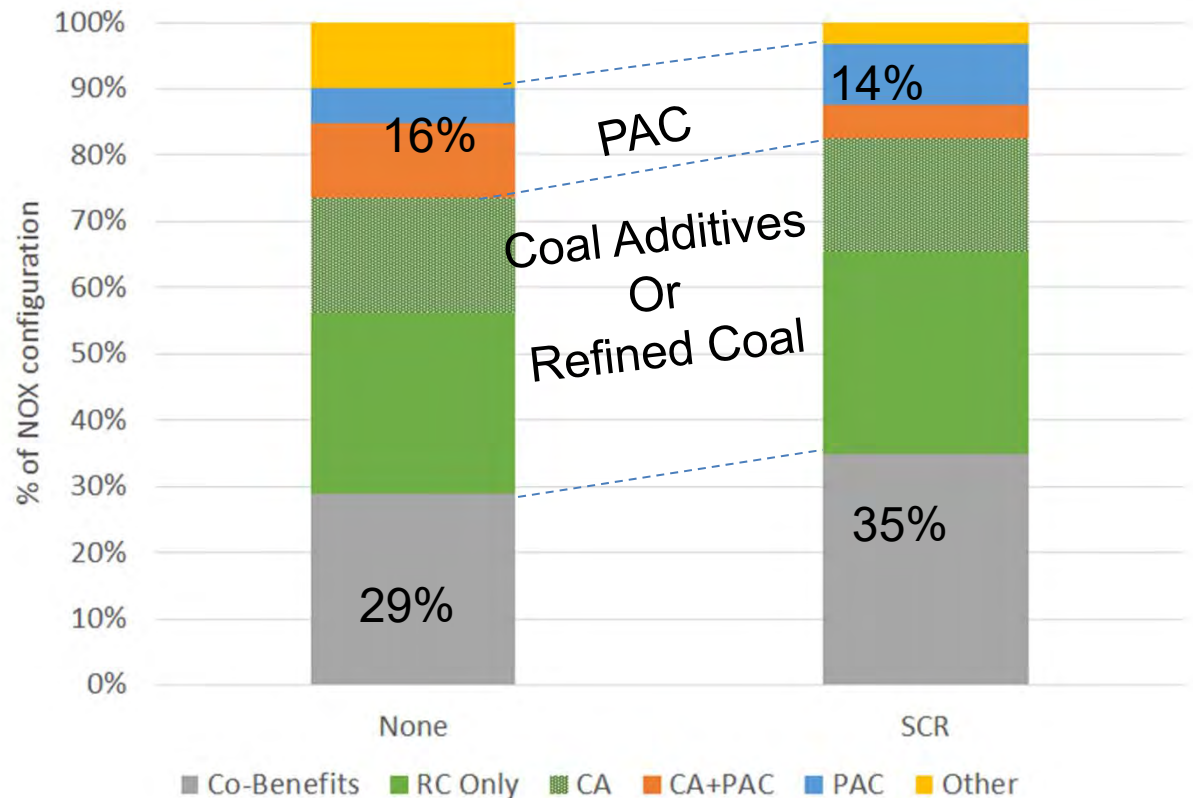
## DSI-ACI Co-Treatment Considerations

- Potential for SO<sub>3</sub> reduction with DSI
  - **improved** Hg capture with PAC
- High-temperature injection of DSI sorbent could **reduce oxidation of Hg** in flue gas by removing halogens too soon
- Sodium sorbents can produce NO<sub>2</sub>
  - can **reduce effectiveness of PAC** for Hg capture
    - Doesn't happen at every DSI installation: reaction kinetics, type of particulate control device (ESP vs. FF), and baseline NO<sub>x</sub> levels are important factors
- Particulate Control impacts
  - Loading increased – also impacts ash handling requirements
  - Resistivity impacted ( ↑Calcium, ↓ Sodium)
- Choose DSI sorbent and injection location carefully

# SOx and NOx Controls: Influence on Mercury Controls: Bituminous-Fired Units



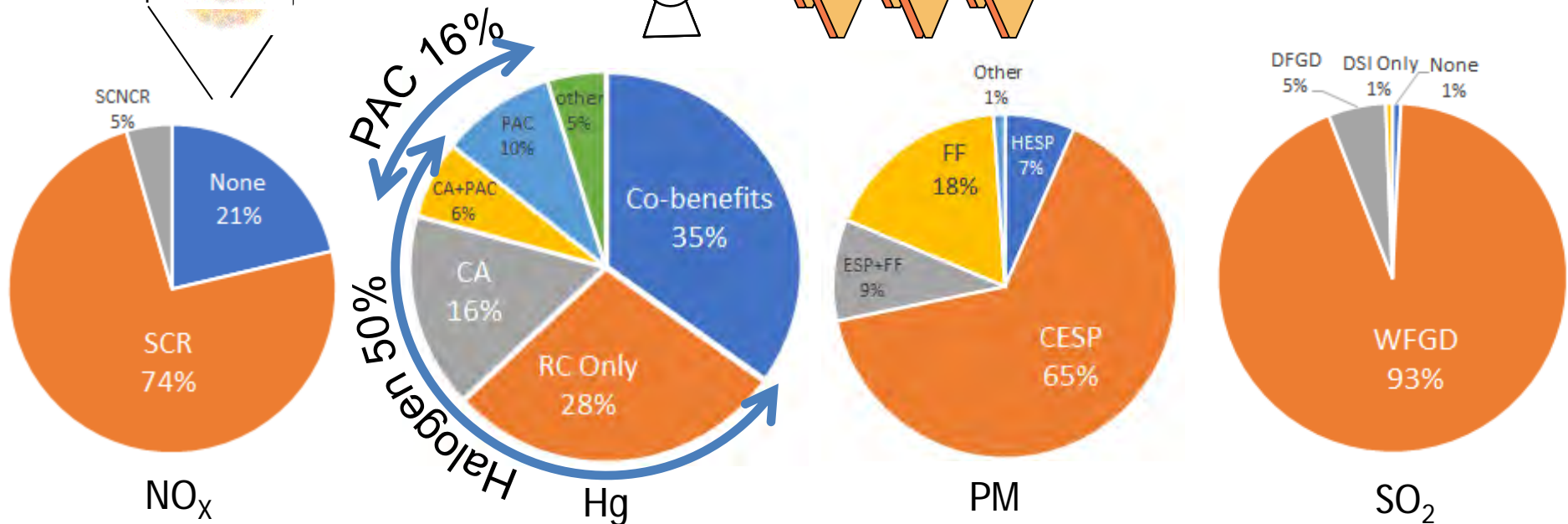
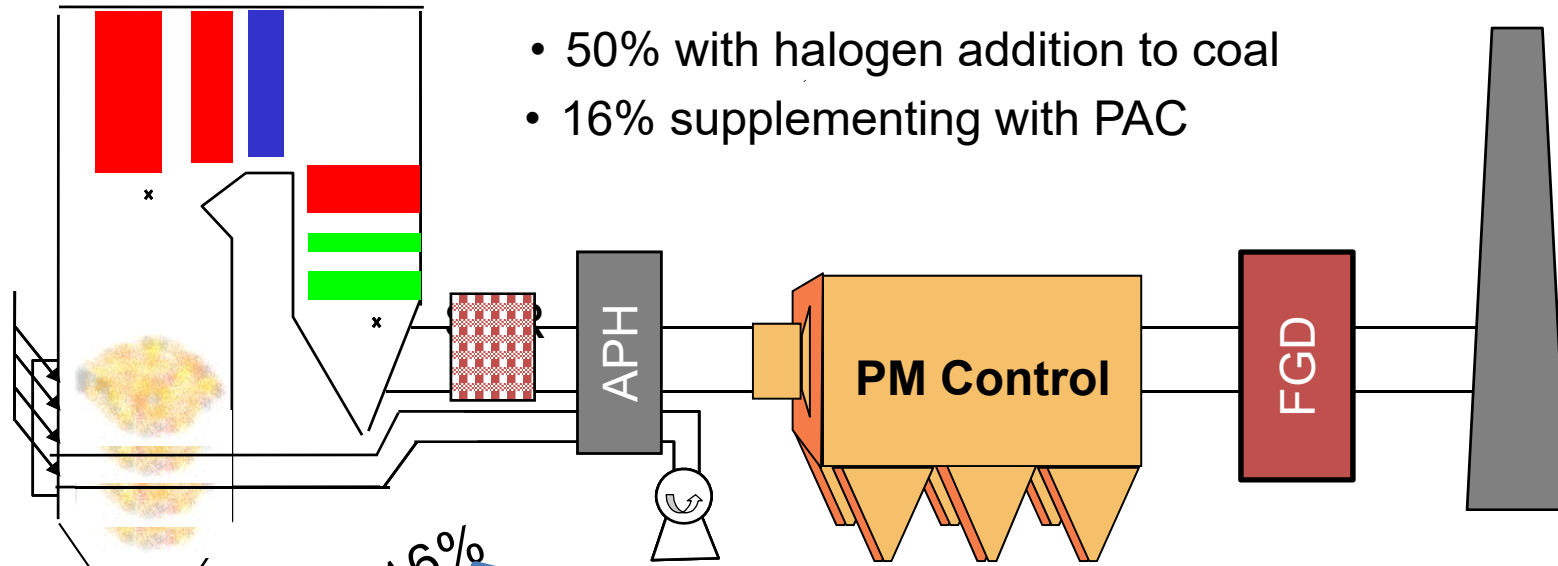
Bituminous-Fired Units



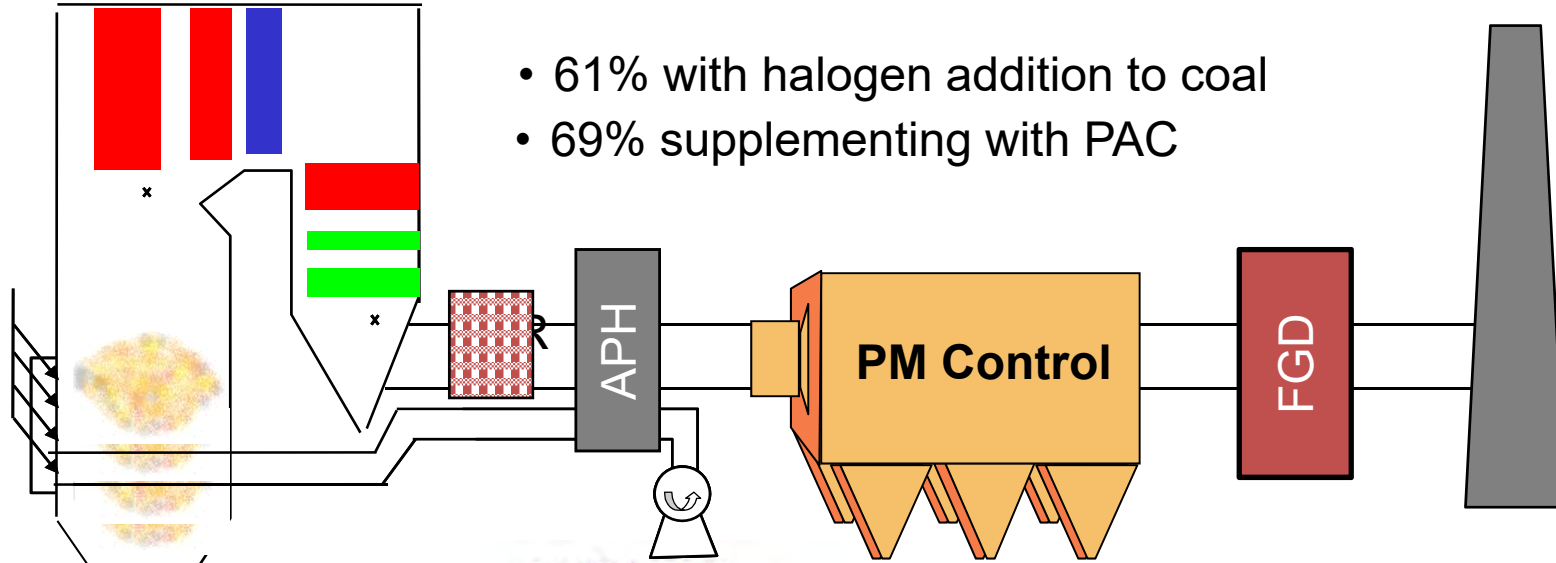
- On average, an installed SCR has a minor impact on the fraction of bituminous units with SCRs reporting using PAC to supplement Hg compliance
  - Data on the amount of PAC or coal additives is not reported
- Larger fraction of SCR + WFGD units relying on co-benefits alone compared to WFGD alone

# Mercury Controls Strategies: Bituminous Plants

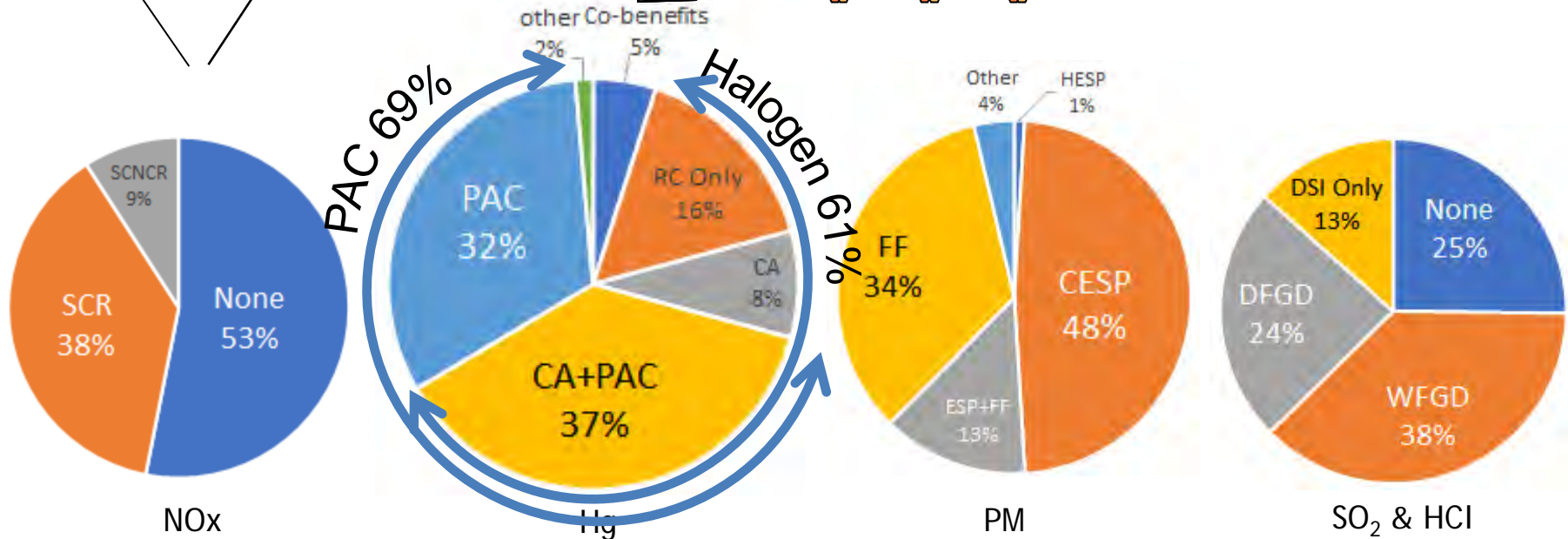
- 50% with halogen addition to coal
- 16% supplementing with PAC



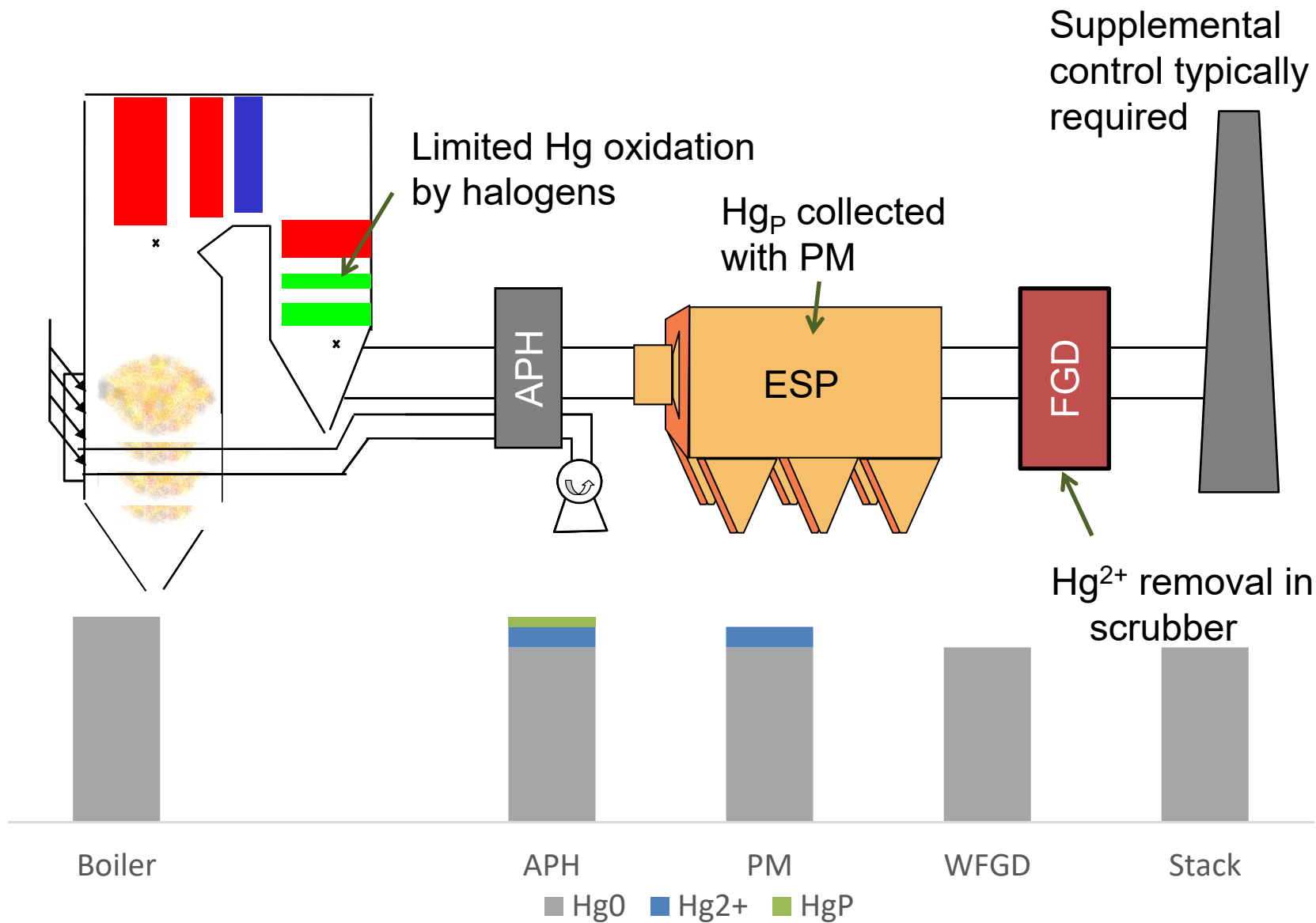
# APC Configurations: Subbituminous Plants



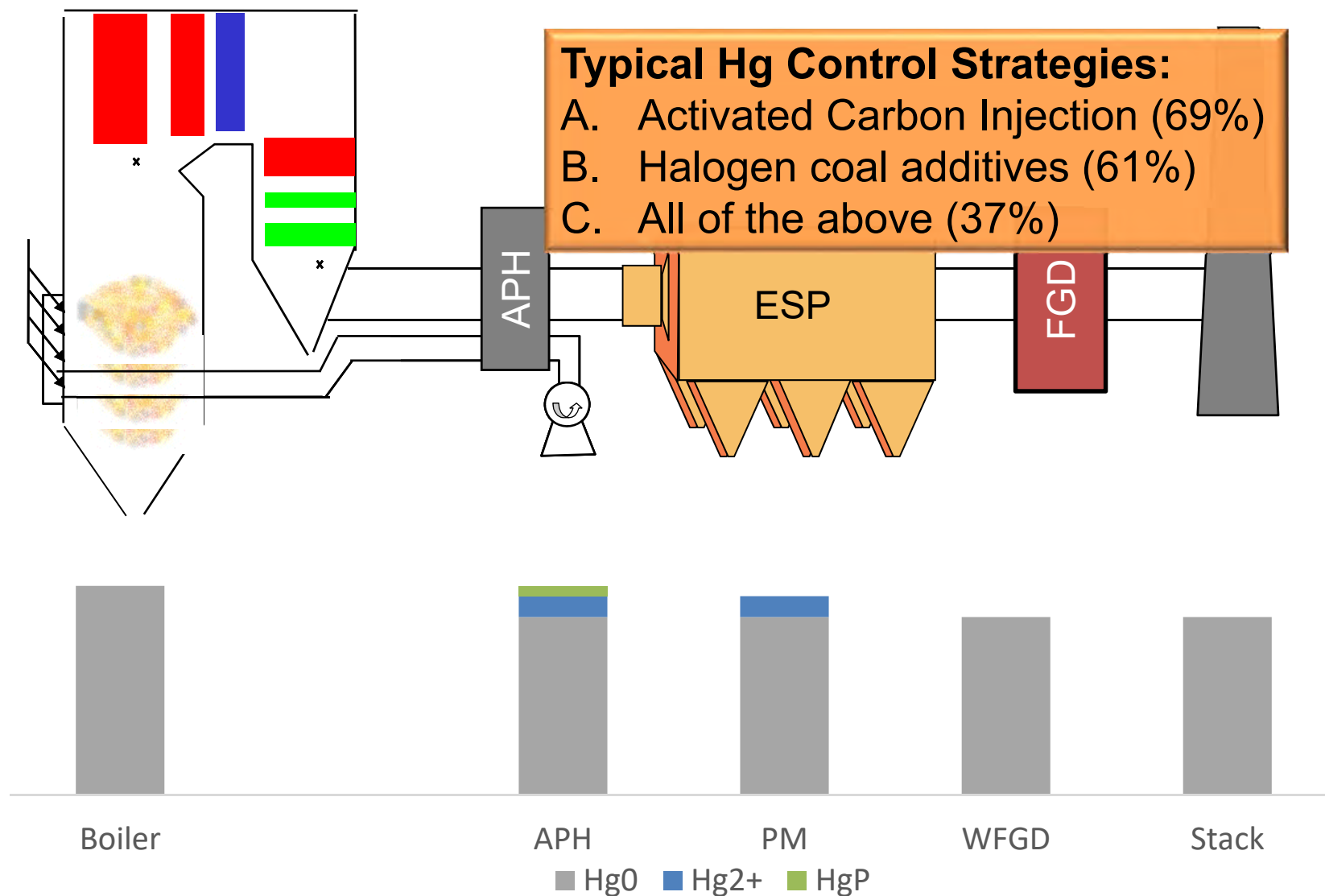
- 61% with halogen addition to coal
- 69% supplementing with PAC



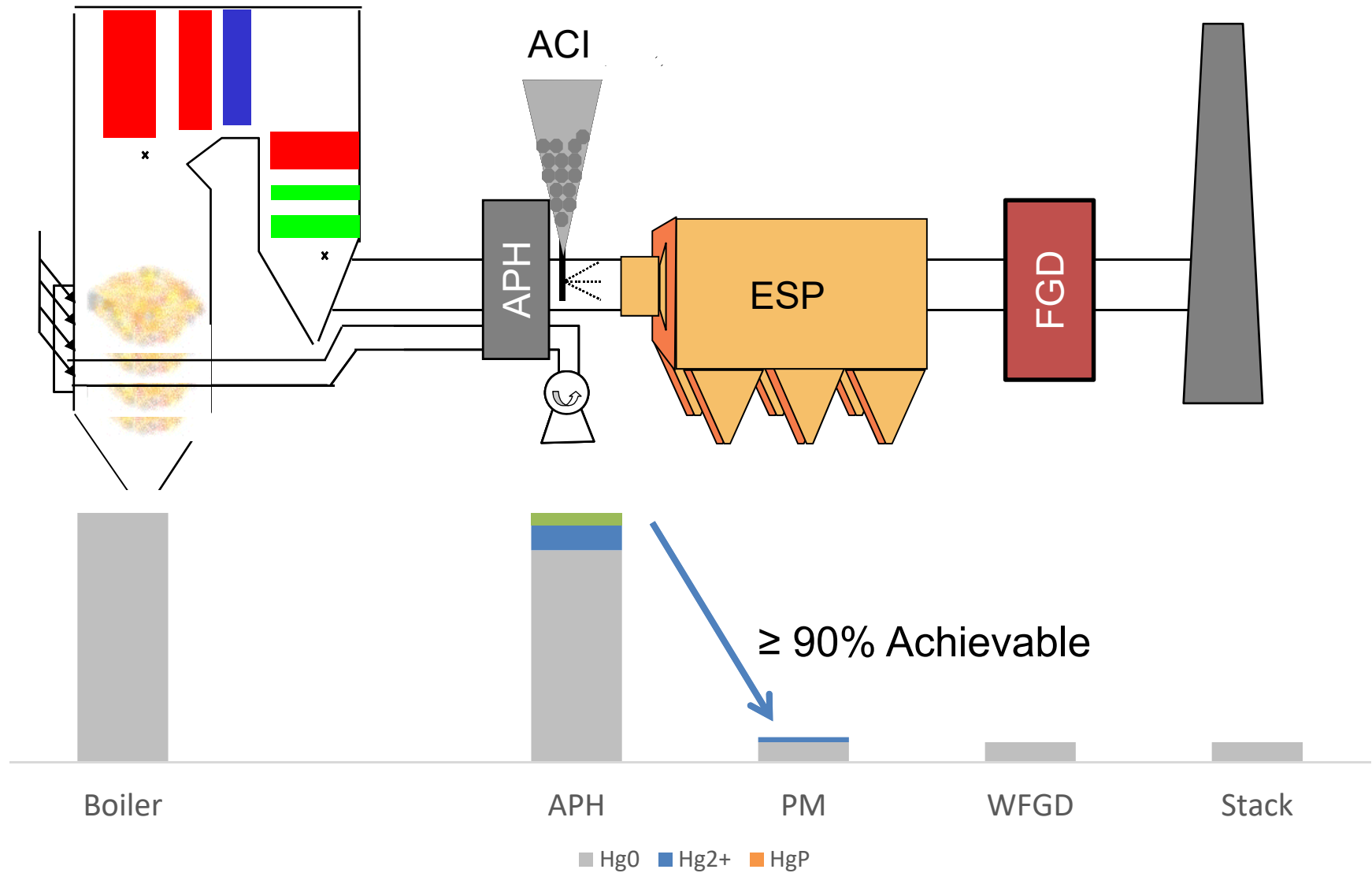
# Subbituminous-Fired Plant with "Typical" APC



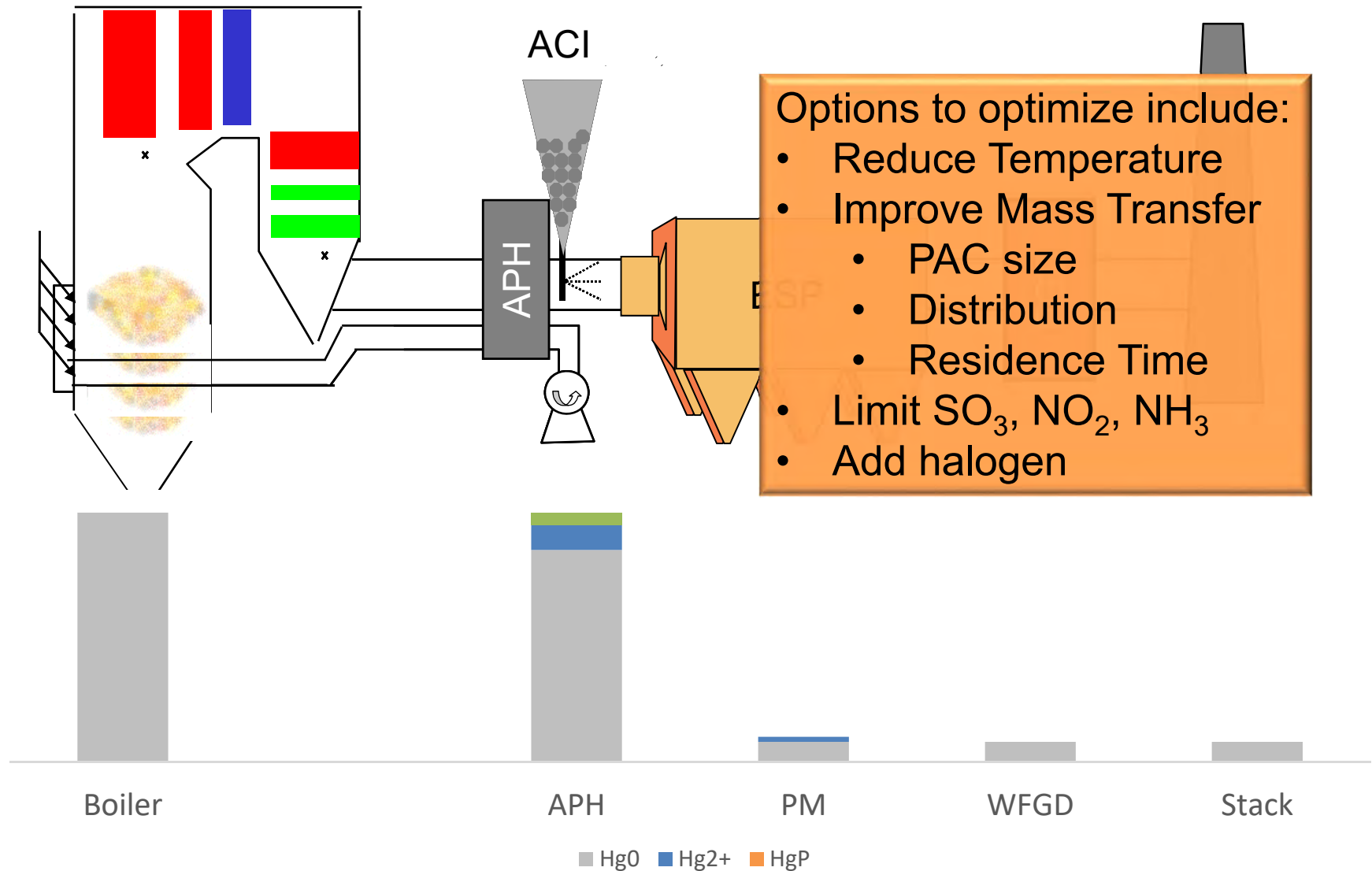
# Subbituminous-Fired Plant with "Typical" APC



# Subbituminous-Fired with "Typical" APC Activated Carbon Injection (ACI)



# Subbituminous-Fired with “Typical” APC Activated Carbon Injection (ACI)

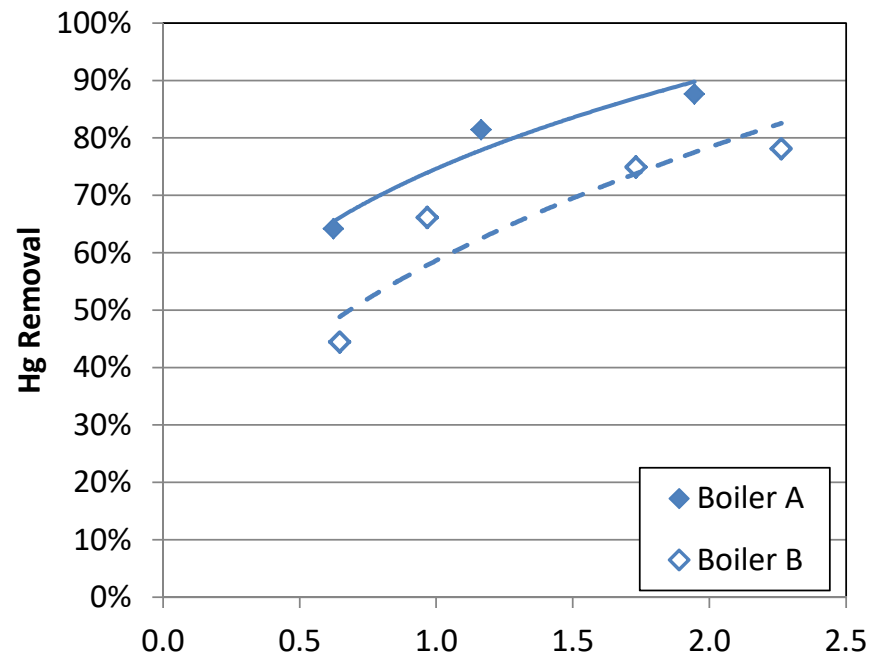


# Mass Transfer Important for ESPs

Powdered Activated Carbon (PAC) usage with an ESP can be reduced by improving mass transfer to the PAC

## Option

- Place lances for longer residence time and optimal distribution
- Reduce PAC particle size



Example: Two boilers burning similar PRB coal  
Testing of brominated PAC on both units at 315-320°F

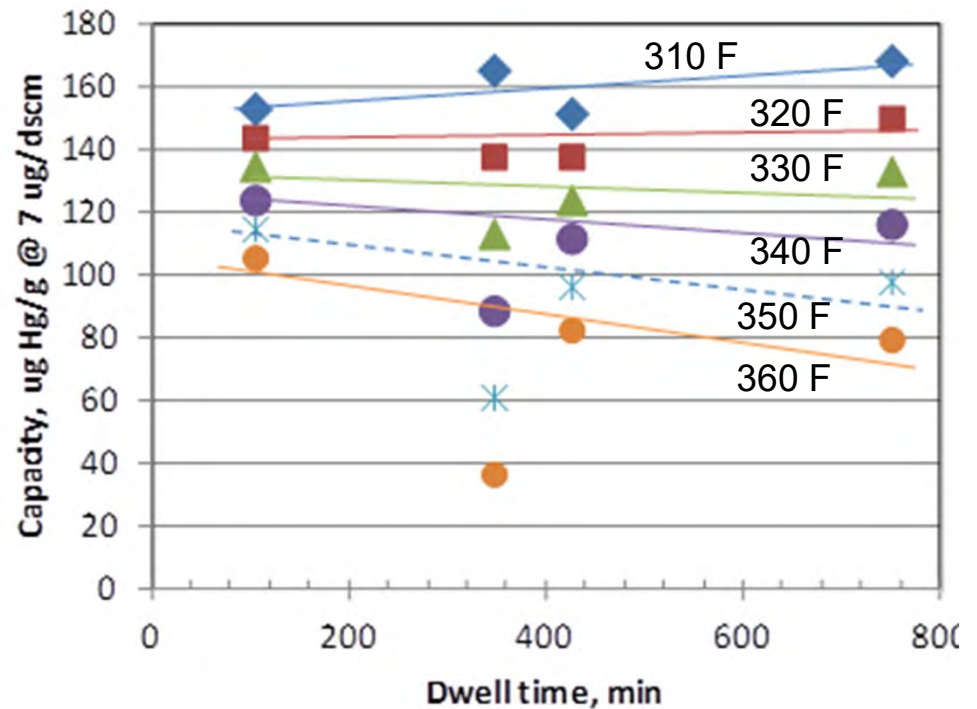
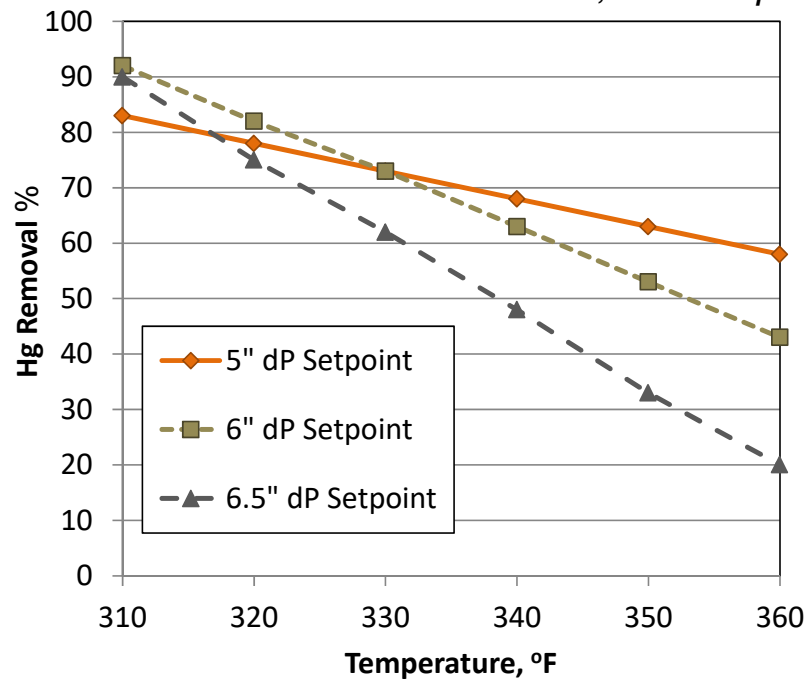
Boiler B has short residence time between APH and ESP inlet AND Chevron-style inlets

Boiler A has longer duct residence time

# Getting the Most Out of ACI with Fabric Filters

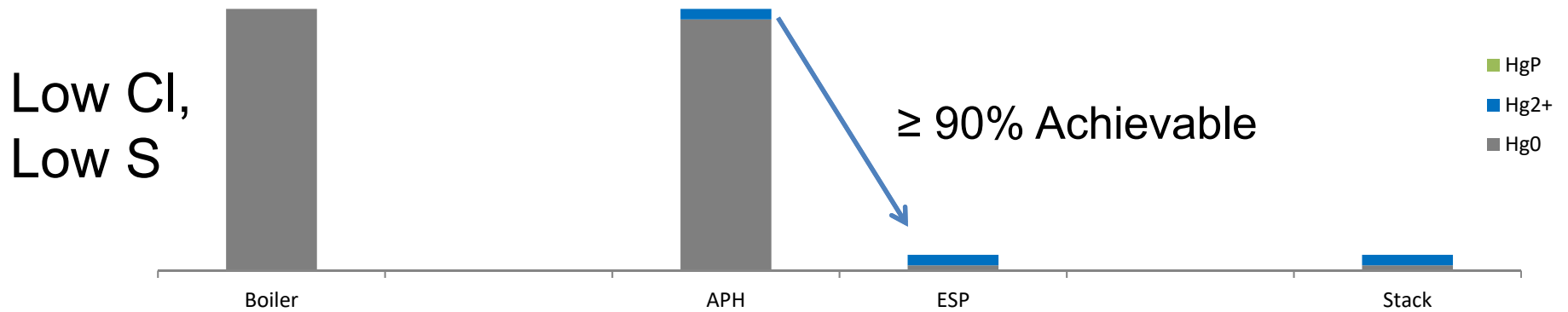
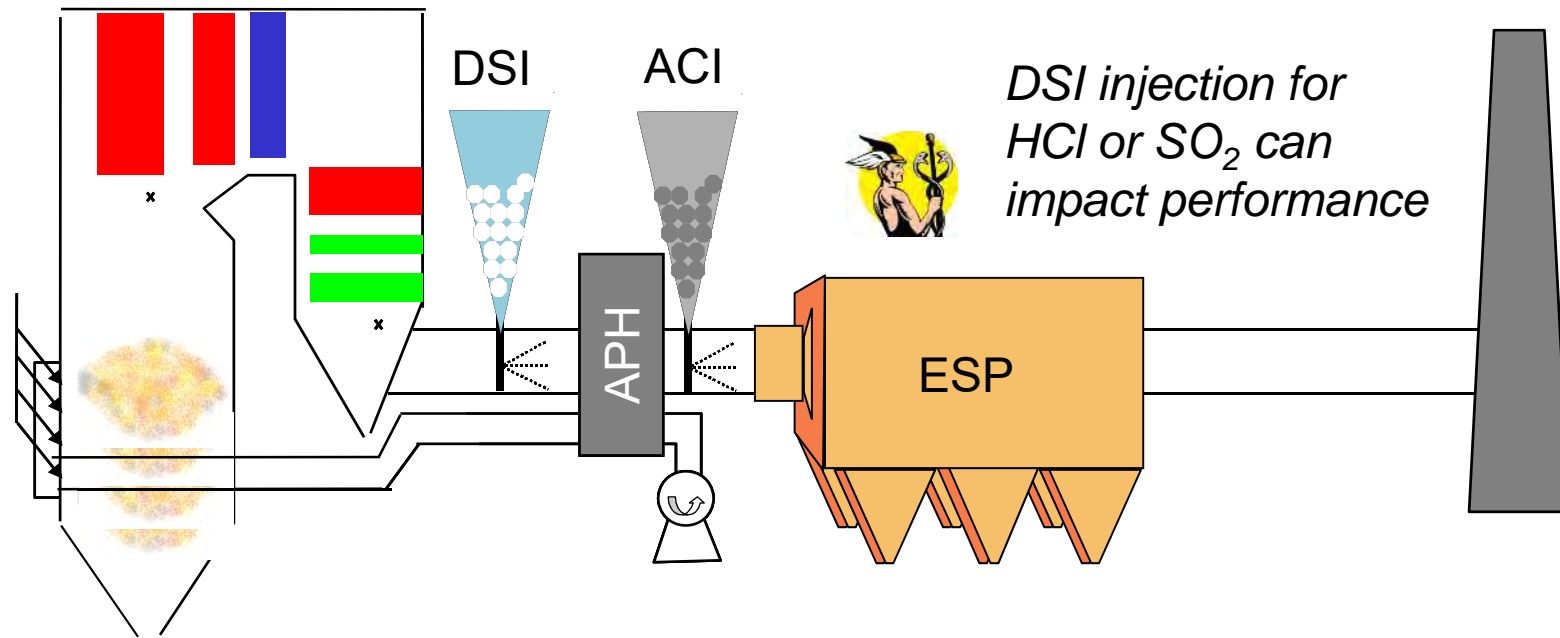
- Lower temperature at the particulate control device
- Provide sufficient contact time
- Clean bags before increasing temperature

Source: Derenne and Stewart, Final Report



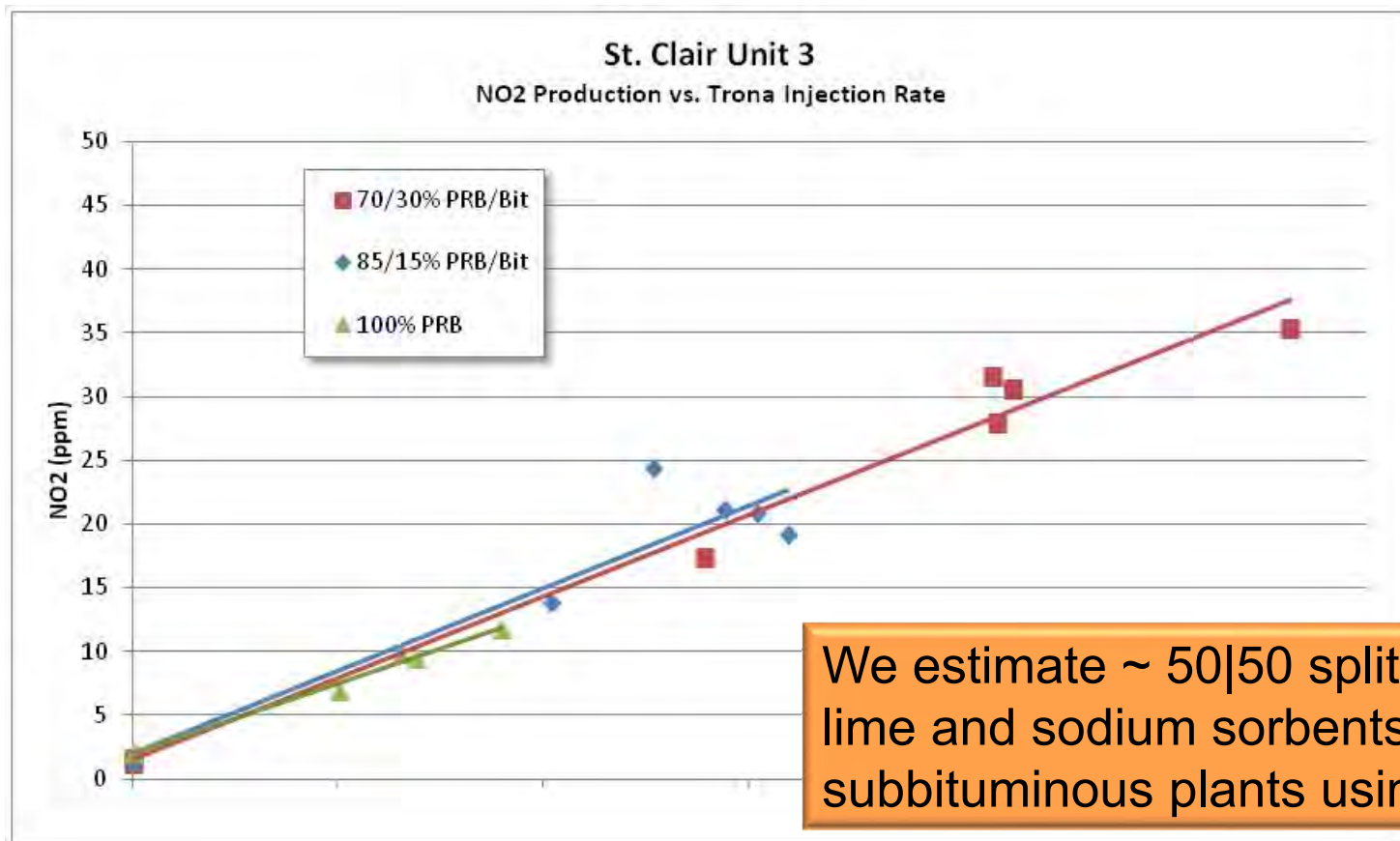
Presque Isle TOXECON fabric filter, 1 lb/MMacf non-brominated PAC

# Mercury and HCl Control Subbituminous Coals



# Potential DSI Challenges: NO<sub>2</sub> Production with Trona Injection

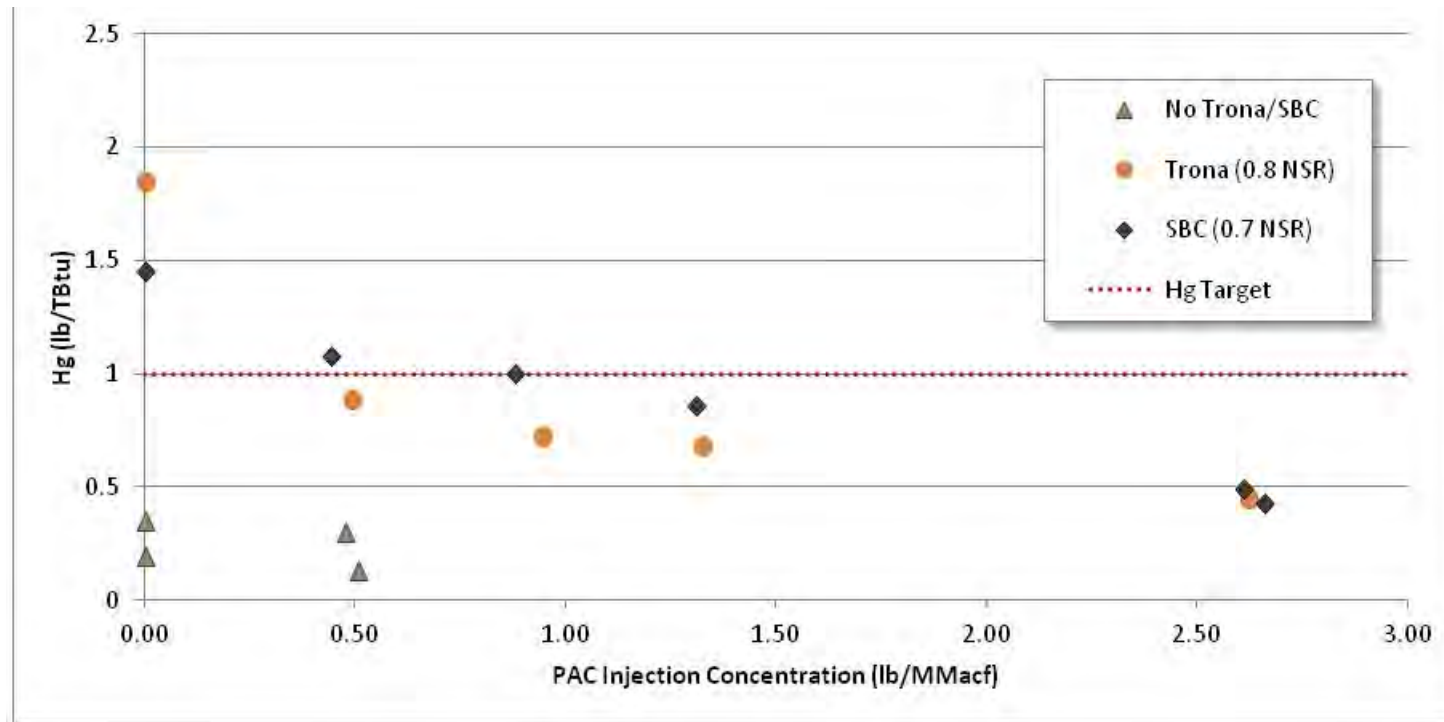
- Sodium-based DSI can increase flue gas NO<sub>2</sub>
- Trona or SBC injection increased stack NO<sub>2</sub> to as much as 35 ppmv



Results from Testing at DTE St. Clair Unit 3, W. Rogers, EUEC 2013

# Impacts of NO<sub>2</sub> on PAC

- NO<sub>2</sub> will interfere with Hg removal by PAC



Non-brominated PAC injected downstream of air preheater and trona or sodium bicarbonate injected upstream of air preheater for HCl control

Results from Testing at DTE St. Clair Unit 3, W. Rogers, EUEC 2013

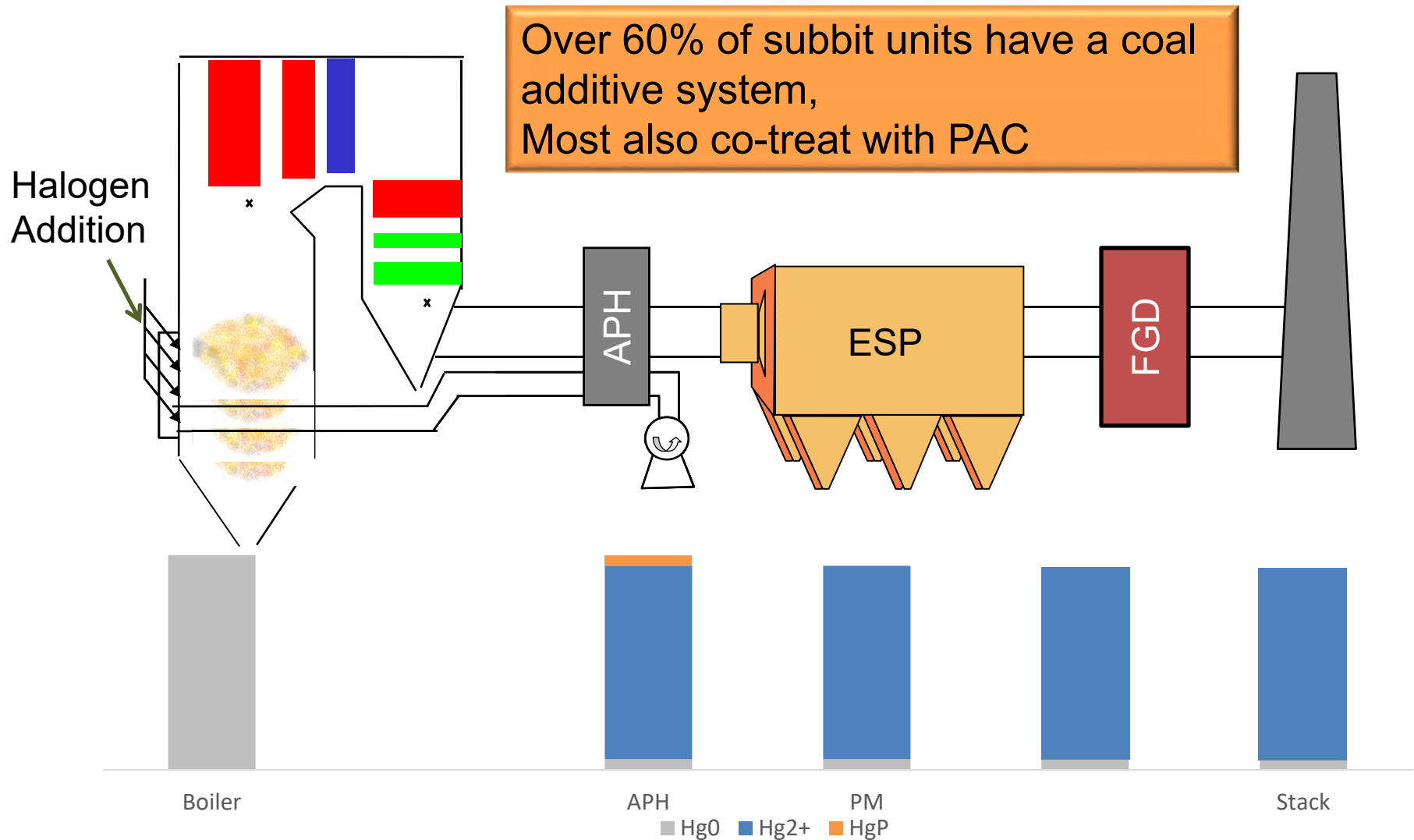
# Increase Native Capture: Halogen Addition

- Adding halogens increases oxidized Hg:
  - Increase effectiveness of some kinds of activated carbon for Hg capture
  - Increase capture of Hg in SO<sub>2</sub> scrubber
- Potential balance-of-plant impacts:
  - Possible increased corrosion
  - Halogens build up in wet scrubber liquor
    - ▶ Average Cl removals for wet FGDs (2010 ICR): 81% for subbituminous, 97% for bituminous
    - ▶ Removal of Br at Plant Miller wet FGD: 94-96% (Dombrowski et al., 2008)
- Bromine treatment may contribute to formation of additional regulated disinfectant byproducts (DBPs) at drinking water treatment plants downstream of FGD wastewater discharge

## Emerging Issue: Bromine in Scrubber Effluent

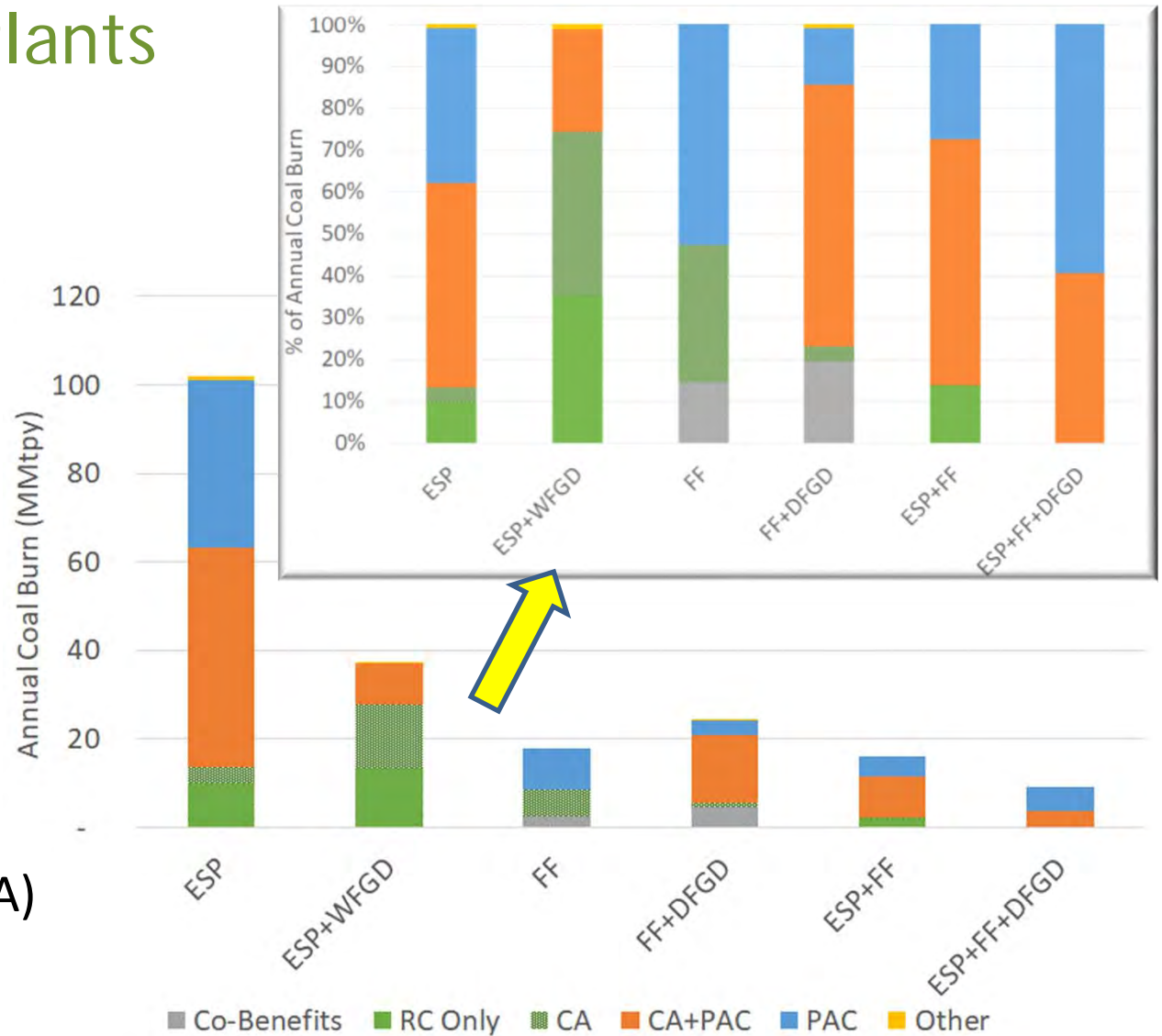
- The US EPA has established limits for eleven DBPs in drinking water after determining that some DBPs are associated with increased cancer rates. These include four trihalomethanes (THMs), five trihalomethanes (HAAs), bromate, and chlorite.
- Increased use of bromine for mercury control has heightened awareness of potential risk associated with bromine-containing effluent discharged into waterways
  - Example: There are > 6,000 public water systems in North Carolina alone

# Subbituminous-Fired Plant, "Typical" APC & Coal Additives (Halogen Injection)



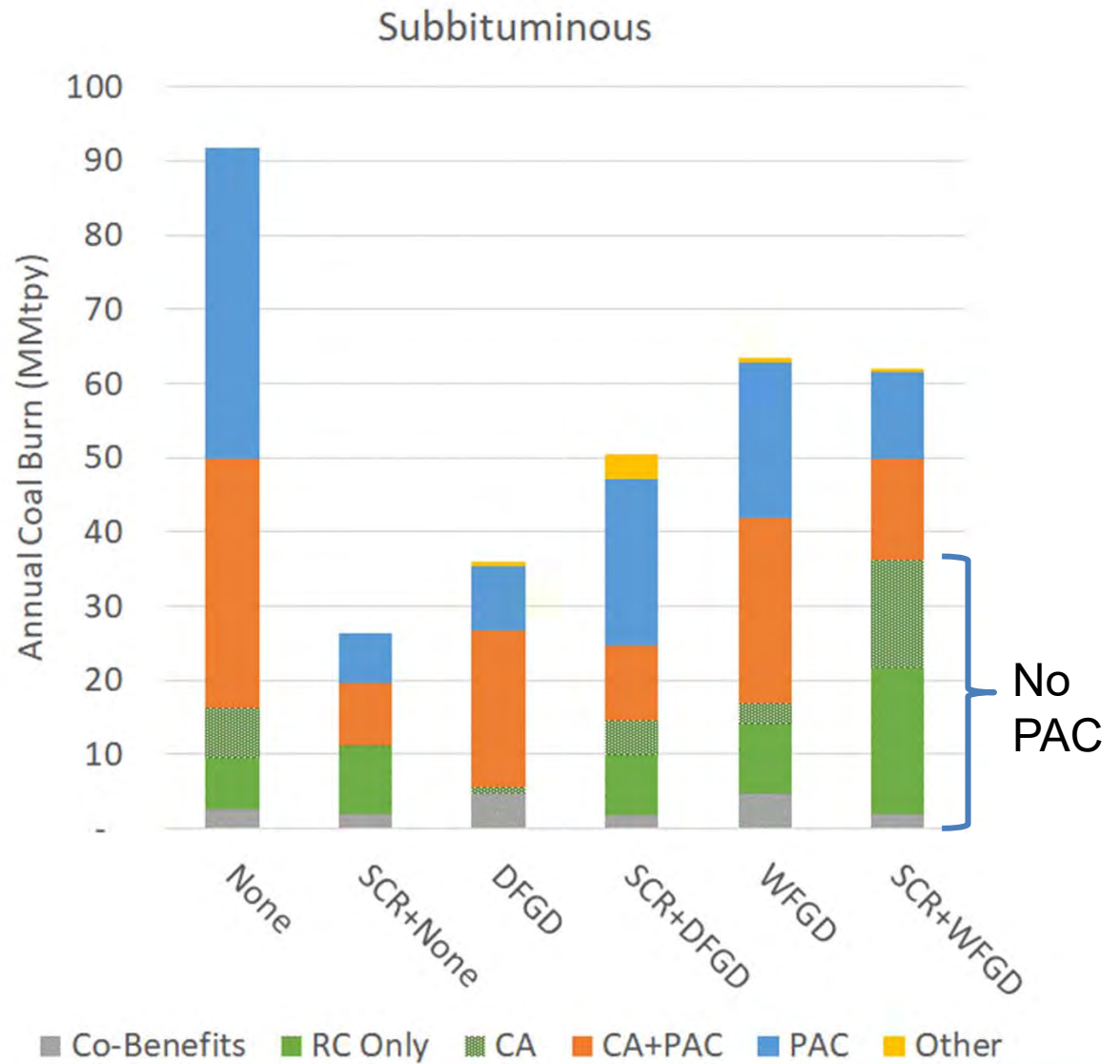
# Impact of Particulate and SOx Controls on Mercury Controls (No SCR or SNCR): Subbituminous Plants

- Only ~ 20% can meet compliance with coal additives alone
- Most units (typ > 80%) require some PAC to meet compliance
- Many use a combination of PAC and coal additives (CA)

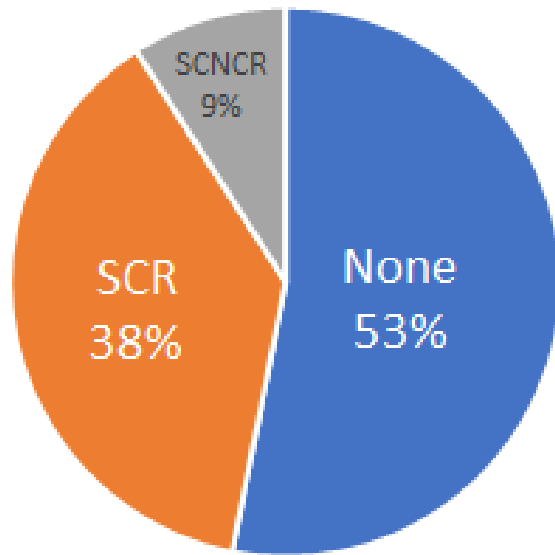


# Impact of SCR + SOx Controls on Mercury Controls

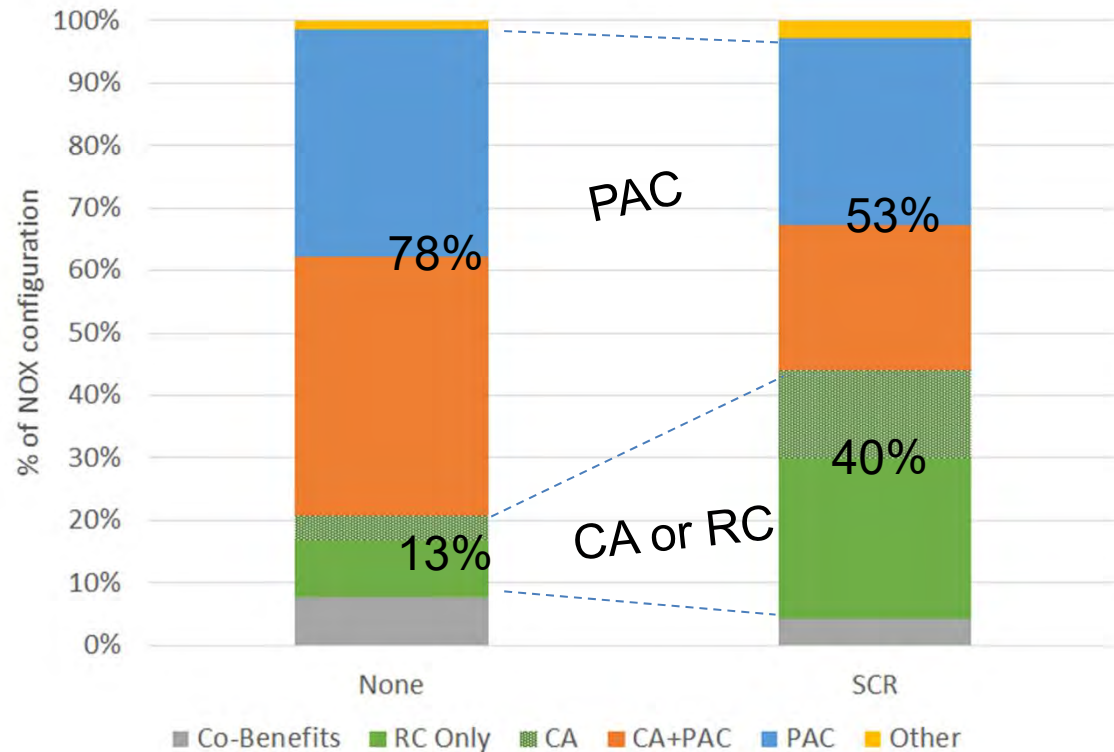
- On average, subbit plants rely on PAC to meet compliance
- Exception: 58% of subbit units with SCR + WFGD meet compliance without any PAC



# Impact of SCR on Mercury Controls

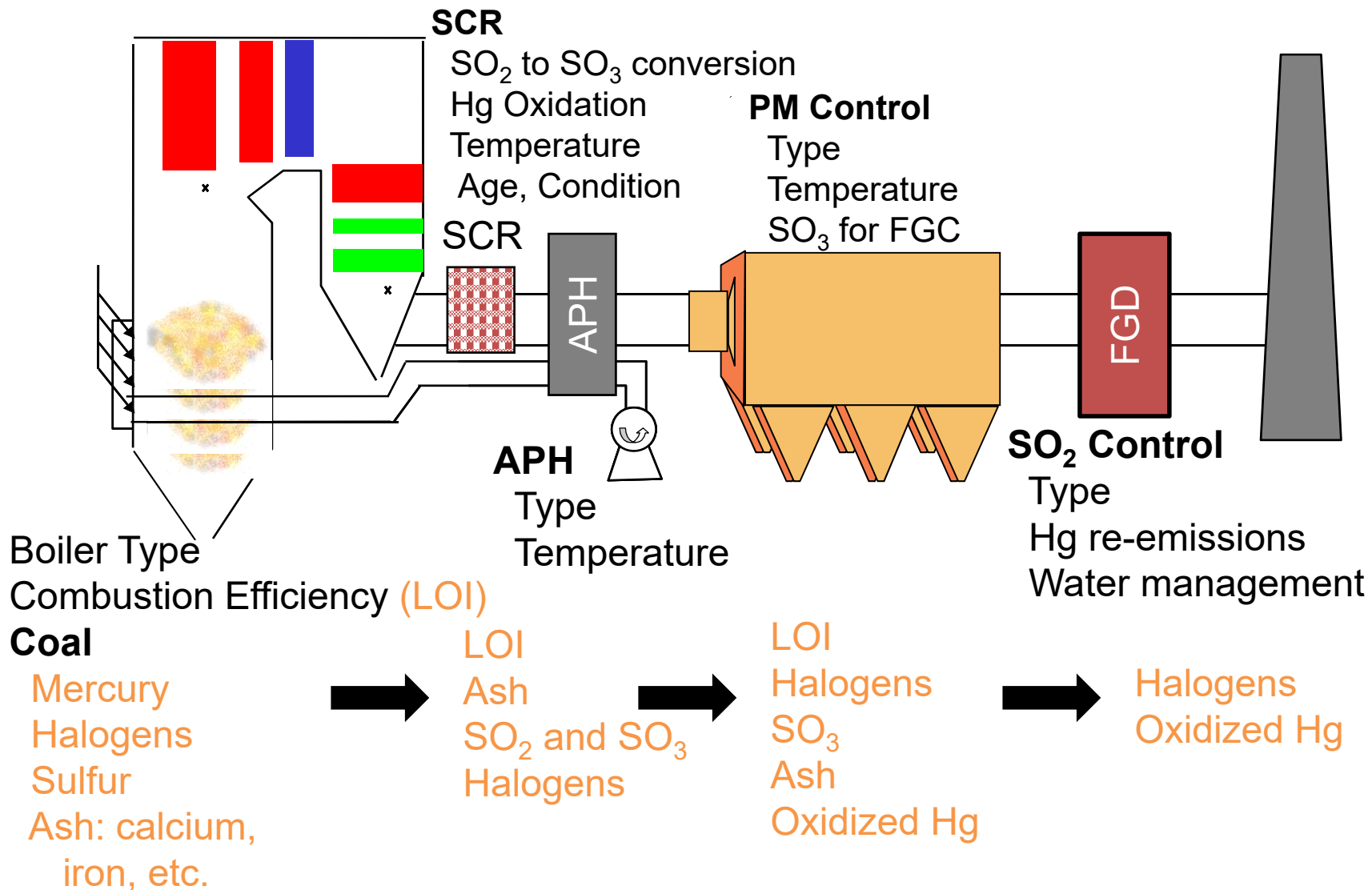


Subbituminous Coal



- Units with installed SCRs are **three times** more likely to report using coal additives alone (no PAC) for Hg oxidation than those without SCRs
  - Data on the amount of PAC or coal additives is not reported

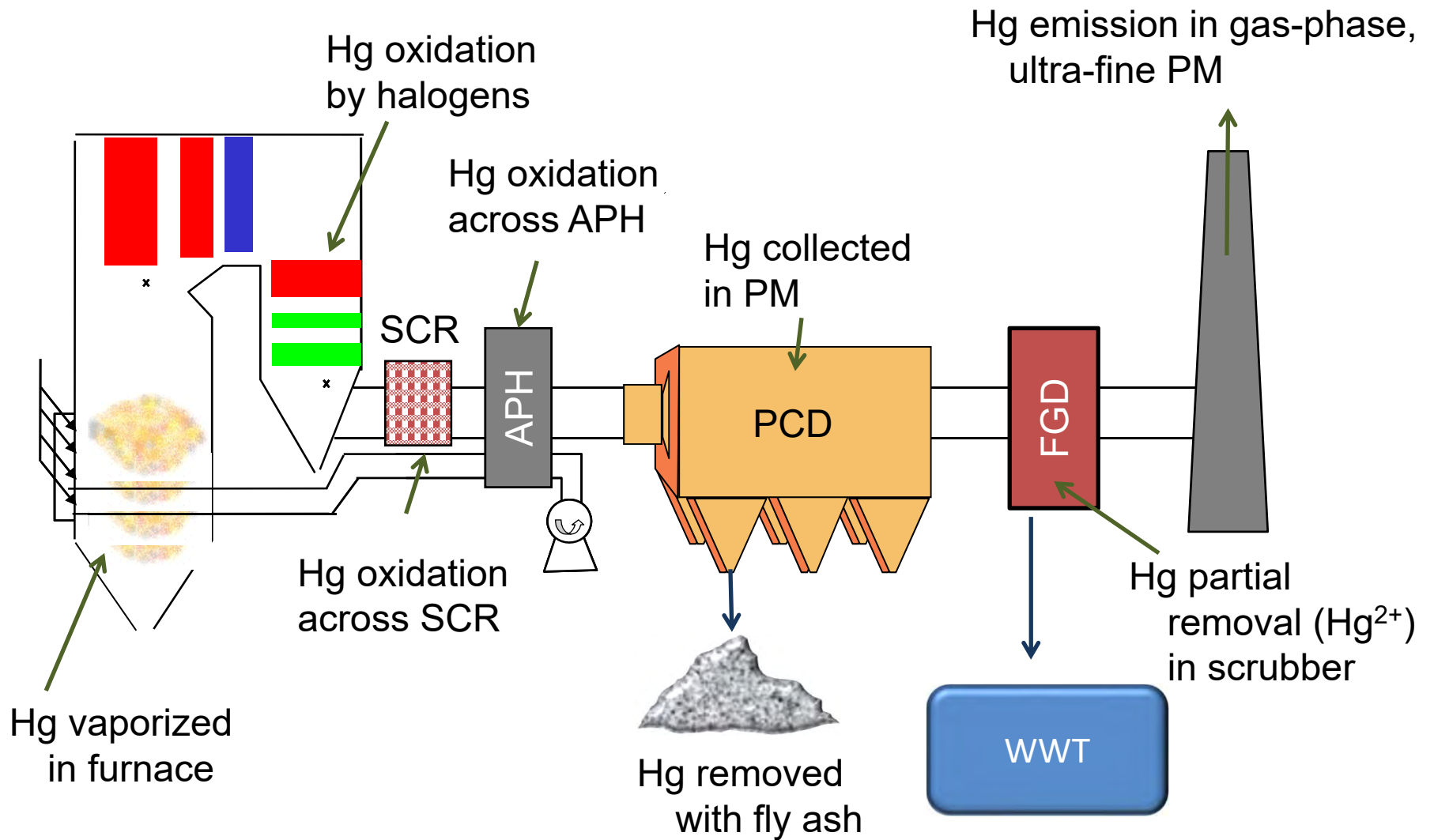
# Review: What's Important for Mercury Control



# Primary Mercury Control Strategies

- Co-Benefits: no mercury-specific controls
  - More common on scrubbed plants firing higher halogen coal
- Coal Additives: Increase halogen content in coal
  - increase fraction of oxidized mercury
  - More common as a stand-alone option on scrubbed plants
- Activated carbon injection: increase concentration of particles that can adsorb mercury
  - Common on plants firing low-halogen fuel

# Fate of Hg in Coal-Fired Boilers



# Compliance Strategies for Mercury

- 80 to >90% control at the stack to meet MATS emission limits required for most units
- MATS limits achievable with ACI or ACI + coal additives on most subbituminous units if SO<sub>3</sub> flue gas conditioning (FGC) is eliminated
- For units with SCR/FGD:
  - Provide sufficient halogens to oxidize the Hg
  - Minimize re-emission of Hg<sup>0</sup> from wet FGD
  - Use ACI as needed for trim
- MATS limits may be challenging on units with higher sulfur coals. Year round compliance may require SO<sub>3</sub> mitigation and careful WFGD re-emissions management

THANK YOU!

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